

الجمهورية الجزائرية الديمقراطية الشعبية

## République Algérienne Démocratique et Populaire

وزارة التعليم العالي والبحث العلمي

Ministère de l'Enseignement Supérieur et de la Recherche Scientifique



جامعة الإخوة منتوري قسنطينة I  
Frères Mentouri Constantine I University  
Université Frères Mentouri Constantine I

Faculté des Sciences de la Nature et de la Vie  
Département de Biologie Animal

كلية علوم الطبيعة والحياة  
قسم بيولوجيا الحيوان

Mémoire présenté en vue de l'obtention du diplôme de Master

Domaine : Sciences de la Nature et de la Vie

Filière : sciences biologiques

Spécialité : Immunologie cellulaire et moléculaire

N° d'ordre :

N° de série :

Intitulé :

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**Biological activities of *Artemisia Judaica* and *Hyacinthoides lingulata***

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Présenté par : Aissaoui Roumaïssa  
Messili Dounya  
Taleb Oussama Ramy

Le 06/06/2022

Jury d'évaluation :

**Promotrice :** RAMLI Iman (Grade MAA - Université Frères Mentouri, Constantine 1).

**Examineur 1 :** HADDAD Souad (Grade MAA - Université Frères Mentouri, Constantine 1).

**Examineur 2 :** B ENLATRECHE Moufida (Grade MAA - Université Frères Mentouri, Constantine 1).

Année Universitaire 2021 - 2022



## *Acknowledgement*

we would like to acknowledge and give our warmest thanks to our supervisor Mrs **RAMLI. IMAN** who made this work possible her guidance and advice carried us through all the stages of writing this project our Thanks also go to the jury members, **Ms BENLATRECHE Moufida** and **Ms HADAD Souad** to accept our invitation to discuss my dissertation, we would like to express our heartfelt gratitude to CRBT laboratory for give us allowance to excess to laboratory to do our experimental part, we greatly appreciate Lynda for her time and assistance.

We would also like to give special thanks for our families and friends as a whole for their continuous support and understanding when under taking our research and writing our thesis.

Finally, we would like to thank God, for letting us through all the difficulties, you are the one how let us finish our degree, we will keep on trusting you for our future.

*I dedicate my work to my mother, a strong and gentle soul who taught me to trust in Allah and that so much could be done with so little, and to my father for his support and encouragement, and their day and night prayers*

*to my brother Rafik and my sister Houda*

*To "Angels Team" Thank you for always believing in me and for the long lunches and lots of giggling.*

*To the adventurer "vanlife" who always gives me advice and time as I need it, and to my best friend Achwak, Bouchra, souheyr, Insaf, and Akram, who push me to see the potential in the ordinary and who always support me.*

*To Ms. Bouaziz Samia, who welcomed me with open arms, taught and supported me, and provided me with a lifetime of wonderful memories.*

*To science for enlightenment.*

*To everyone in Microsoft Office who has a red line under their name.*

---

*Dounya*

**I dedicate this message to myself, who stayed up late at night and struggled to succeed to my dad who worked hard to get to this point, to my mom who supported me with motivational words and phrases, and to my brother who didn't.**

**Forgive me for anything, whatever I ask,  
and for my friends Hadeel Bushalgam, Nour Lakash  
and Hanan Ben Sassi who supported me morally and helped me in the most difficult  
circumstances.**

**Roumaysa**

### *Dedication*

This thesis is dedicated to my mother who passed away, Saba Zineb, and my father, Ahmed. Without their endless love and encouragement, I would never have been able to complete my graduate studies. I love you both and I appreciate everything that you have done for me.

Thank you, mom, for everything

This thesis is also dedicated to my sisters and my brothers, Sara, Rayen, Seif, Souheil they were there for me throughout this process and gave me lots of support.

**Ramy**

# *Table of content*

## *Acknowledgments*

<b>General Introduction</b> .....	<b>1</b>
<b>I. Oxidative Stress and Disorders</b> .....	<b>3</b>
I. Oxidative Stress .....	4
I.1. Introduction.....	4
I.2. Free Radicals .....	<b>4</b>
I.3. Reactive Species .....	5
I.3.1. Reactive Oxygen Species .....	7
I.3.2. Types of ROS.....	9
I.3.3. Sources of ROS .....	11
I.3.3.1. Reactive Nitrogen Species (RNS) .....	11
I.3.3.2. Reactive Sulfur Species (RSS) .....	11
I.3.3.3. Reactive Carbonyl Species (RCS) .....	11
I.4. Inflammation .....	12
I.4.1. Causes of Inflammation .....	12
I.4.2. Types of Inflammation .....	13
I.4.2.1. Acute Inflammation .....	13
I.4.2.1. Chronic Inflammation .....	14
I.4.3. Differentiation Points between Acute and Chronic Inflammation .....	15
I.4.4. The Relationship between Oxidative Stress and Inflammation .....	15
I.4.5. Oxidative Stress and Disease .....	15
I.4.5.1. Inflammation .....	16
I.4.5.2. Cardiovascular Disease .....	16
I.4.5.3. Neurodegenerative disease .....	16
I.4.5.4. Rheumatoid Arthritis .....	17
I.4.5.5. Ischemia/Reperfusion Injury .....	17
I.4.5.6. Diabetic .....	17
I.4.5.7. Aging .....	17
I.4.5.8. Cancer .....	18
<b>II. Secondary Metabolites and Their Biological Activities</b> .....	<b>19</b>
II.1. Polyphenols .....	20
II.1.1. Non-flavonoid Polyphenols .....	21
II.1.1.1. Simple Phenols .....	22

II.1.1.2. Phenolic Acids .....	22
II.1.1.3. Flavonoids .....	22
II.1.1.4. The Isoflavonoids .....	24
II.1.1.5. Anthocyanins .....	25
II.1.1.6. Tannins .....	25
II.2. Biological Properties of Polyphenols .....	25
II.2.1. Antioxidant Activity .....	26
II.2.2. Cardioprotection Activity .....	26
II.2.3. Anticancer Activity .....	27
II.2.4. Anti-Inflammation Activity .....	27
II.2.5. Anti-aging Effect .....	27
II.3. Terpenes .....	28
II.4. Alkaloids .....	29
II.5. Molecules Identified from Scilla and Artemisia and Their Biological Activity . .....	29
II.5.1. From the Genus Hyacinthoideae.....	29
II.5.2. From the Genus Artemisia .....	30
<b>III. Generalities about Hyacinthoides Lingulata and Artemisia Judaica .....</b>	<b>31</b>
III.1. Studied Plants .....	32
III.1.1. Artemisia Judaica .....	32
III.1.2. Chemical Composition .....	33
III.1.3. Biological and Therapeutic Activity .....	33
III.2. Hyacinthoides Lingulata .....	34
III.2.1. Biological and Therapeutic Activity of Hyacinthoides Lingulata .....	35
<b>IV. Materials and Methods .....</b>	<b>37</b>
IV.1. Materials .....	38
IV.1.1. Plant material and crude extract preparation.....	38
IV.2. methods .....	38
IV.2.1. Antioxidant Activity .....	39



<b>IV.2.2. DPPH Radical Scavenging Activity Assay</b> .....	39
<b>IV.2.3. Reducing Power Assay</b> .....	40
<b>IV.2.4. ABTS Radical Cation Scavenging Activity Assay</b> .....	40
<b>IV.2.5. Phenanthroline Assay</b> .....	41
<b>IV.2.6. Cupric Reducing Antioxidant Capacity</b> .....	41
<b>IV.2.7. Silver Nanoparticle (SNP) Assay</b> .....	41
<b>IV.3. Anticholinesterase Activity</b> .....	42
<b>IV.4. The <math>\alpha</math>-amylase Inhibition Activity</b> .....	42
<b>V. Results and Discussion</b> .....	43
<b>V.1. antioxidant activity of <i>A. Judaica</i> and <i>H. lingulata</i></b> .....	44
<b>V.2. Cholinesterase Inhibitory Activity</b> .....	44
<b>V.3. Antidiabetic Inhibitory Activity of extracts</b> .....	45
<b>V.4. Discussion</b> .....	46
<b>V.4.1. Antioxidant and antiradical activity</b> .....	46
<b>V.4.2. The Antidiabetic inhibitory activity of extracts</b> .....	46
<b>Conclusion</b> .....	47
<b>References</b> .....	48

# Liste of Figures

<b>Figure 1.</b> Basic concepts and components of oxidative stress.....	4
<b>Figure 2.</b> Effects of oxidative stress on the human body.....	5
<b>Figure 3.</b> Antioxidant/pro-oxidant balance, SOD: superoxide dismutase, CAT: catalase, GSH-Px: glutathione peroxidase, Vit. C/E: vitamin C and E .....	6
<b>Figure 4.</b> Oxidative stress .....	7
<b>Figure 5.</b> Neutralization of a free radical by an antioxidant .....	8
<b>Figure 6.</b> Main Reactive Oxygen Species .....	9
<b>Figure 7.</b> Effects of ROS .....	12
<b>Figure 8.</b> Proresolving mediators in the inflammatory response .....	16
<b>Figure 9.</b> Oxidative stress-induced diseases in humans .....	20
<b>Figure 10.</b> Classification of Polyphenols .....	23
<b>Figure 11.</b> Flavonoid structure .....	25
<b>Figure 12.</b> The skeletons of the isoflavonoids with three-ring structures .....	28
<b>Figure 13.</b> Polyphenols and their biological properties .....	33
<b>Figure 14.</b> <i>Artemisa Judaica</i> .....	35
<b>Figure 15.</b> <i>Hyacinthoides lingulata</i> .....	39
<b>Figure 16.</b> Application of free radical diphenylpicrylhydrazyl (DPPH) to estimate the antioxidant.....	45
<b>Figure 17.</b> $\alpha$ amylase inhibitory activities of <i>A. Judaica</i> .....	46
<b>Figure 18.</b> $\alpha$ amylase inhibitory activities of <i>H. lingulata</i> .....	46.

# Liste of Tables

<b>Table 1.</b> Major Active Oxygen Species.....	5
<b>Table 2.</b> Sources of Reactive oxygen species in the body .....	10
<b>Table 3.</b> Markers correspond to both acute inflammation .....	14
<b>Table 4.</b> Markers correspond to chronic inflammation .....	14
<b>Table 5.</b> Differentiation points between acute and chronic inflammations include .....	15
<b>Table 6.</b> Structure of polyphenol skeletons .....	21
<b>Table 7.</b> Classes of flavonoids .....	23
<b>Table 8.</b> Classification of <i>Artemisia Judaica</i> .....	32
<b>Table 9.</b> Botanical classification of <i>Hyacinthoides lingulata</i> .....	35
<b>Table 10.</b> the antioxidant activity of <i>A. judaica</i> and <i>H. lingulata</i> .....	45
<b>Table 11.</b> Butyrylcholinesterase inhibitory activities of <i>A. judaica</i> and <i>H.lingulata</i> , BChE assay .....	45
<b>Table 12.</b> $\alpha$ amylase inhibitory activities of <i>A. judaica</i> and <i>H.lingulata</i> .....	45

## *Abbreviations & Symbols used*

IO <sub>2</sub>	Oxygène singulier
AlCl <sub>3</sub>	Chlorur of aluminium
BHT	butylated hydroxytoluene
C	Carbon
C <sub>4</sub> H	Cinnamate 4-hydroxylase
Cu <sup>2+</sup>	Ion cuivre
DNA	Deoxyribonucleic acid
DO	Optic Density
DPPH	2, 2-diphényl- 1 –picrylhydrazyl
Fe <sup>2+</sup>	Ions ferreux
FeCl <sub>3</sub>	Chlorure ferric
g	Gramme
g/l	Gramme per litre
H	Hydrogen
H <sub>2</sub> O	Distillate water
H <sub>2</sub> O <sub>2</sub>	Peroxyde d'hydrogène
H <sub>2</sub> SO <sub>4</sub>	Sulfuric Acid
HCl	Chlorhydric Acid
HOCl	Hypochloric Acid
IC <sub>50</sub>	50% inhibition concentration
K <sub>2</sub> CrO <sub>3</sub>	Chrome de potassium
C	carbon
KH <sub>2</sub> PO <sub>4</sub>	Potassium dihydrogen phosphate
M	Molaire
MeOH	Méthanol
mg	Milligramme
ml	Millilitre
mn	Minute
NaOH	Hydroxyd of sodium
nm	Nanometr
NO•	Monoxyd azote
O	Oxygen
O <sub>2</sub> • <sup>-</sup>	Anion superoxyde
OCH <sub>3</sub>	Methoxyl Groupe
OH	Hydroxyl
OH•	Radical hydroxyle
ONOO <sup>-</sup>	Peroxynitrite
R	Radical
ROS	Reactive oxygen spaces
SOD	Superoxyde dismutase
To	Temperature

# *General Introduction*

## General introduction

Medicinal plants are Nature's gift to human beings to help them pursue a disease-free healthy life, since thousands of years ago, people from all cultures use traditional herbal remedies to treat their sicknesses and solve health problems that cause human suffering by using plants directly or indirectly, especially in the plants-rich region we used to see our grandmothers using natural mixtures, as a source of healing (Bora and Sharma., 2011).

Up to this point, it has been observed a huge interest in medicinal crops to extract oils and bioactive compounds to use them as food additives to delay or prevent the growth and development of microorganisms in our medication; additionally, because it is underexplored, this costly healthcare plant has the potential to be a source of new drugs. To accomplish this, an extensive investigation is required as a first step in the creation of a benchmark medical bank that groups the values of each species and their potent effects. (Malik et al.,2019)

The genus *Hyacinthoides* of the "Hyacinthaceae" have numerous systematic, genetic, evolutionary, and ecological interests. From a biochemical and pharmacopoeial standpoint, they are known for their abundant production of triterpenes, including heterosides (scillirosides) with cardiogenic and rat poison activity. Some species contain flavonols, kaempferol, and quercetin. It is thought to be a possible source. (Nishida et al.,2008). a list of medicinal plants that are well-known for treating female infertility, and constipation (van Wyk.,2011) promoting blood circulation, and acting as an anti-inflammatory and analgesic (Nishida et al.,2008).

The small perennial plant *Hyacinthoides lingulata* (Poir.) Rothm. is native to Northern Africa. In Algeria, it is known as 'Becal el Far,' and it is used to treat menopause and gynecological problems (Sabah Chermat and Rachid Gharzouli 2015). Numerous biological studies have shown glycosidase inhibitory (Watson et al. 1997a; Funakoshi-Tago et al.,2011), anti-inflammatory (Hafez Goran et al. 2014; Thomford N 2018), anti-cancer (Bora and Sharma.,2011), and antioxidant activity. (Tan RX, Zheng WF, Tang HQ., 1998).

Natural products and related structures play an important role in pharmaceutical products due to the variety and functionality of secondary metabolites in plant species. The developments of powerful analytical tools in the 21st century such as high-liquid chromatography, nuclear magnetic resonance spectroscopy, mass spectrometry, and the

availability of advanced *in vitro* screening methods greatly expedite the identification and characterization of these natural products, in particular, those of medicinal plants (Thomford N.,2018).

In the same context, the genus *Artemisia* a genus of annual, perennial, and biennial herbs (family of Asteraceae) includes more than 500 species (Bora and Sharma,2011).

*Artemisia* species have a wide range of biological activities, including antimalarial, cytotoxic, antihepatotoxic, antibacterial, antifungal, and antioxidant activity. Artemisinin, the well-known antimalarial drug isolated from the Chinese herb *Artemisia annua*, is one of the most important drugs leads discovered from this genus. Terpenoids, flavonoids, coumarins, caffeoylquinic acids, sterols, and acetylenes are the genus' major phytoconstituents. (Bora and Sharma.,2011).

Furthermore, oxidative stress influences chronic diseases such as cardiovascular disease, diabetes, neurodegenerative diseases, and cancer. Long-term pro-oxidant factor exposure can result in mitochondrial DNA structural defects as well as functional changes in several enzymes and cellular structures, resulting in gene expression abnormalities (Hu et al.,2020). There are several groups of compounds found in plants that have the potential to treat diseases caused by oxidative stress. (Pérez-Torres.,2021)

From this perspective, our research work is focused on the phytochemical analysis of the extract of the plant *Scilla lingulata* and *Artemisia Judaica* and a variety of biological activities.

*Chapter I*

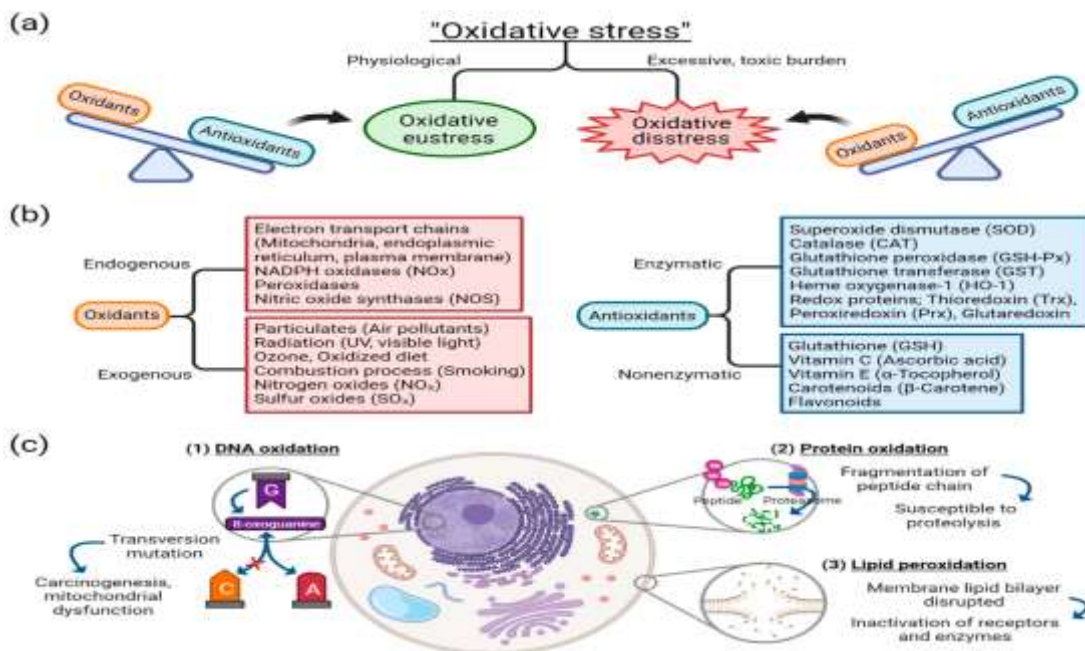
***OXIDATIVE STRESS AND  
DISORDERS***



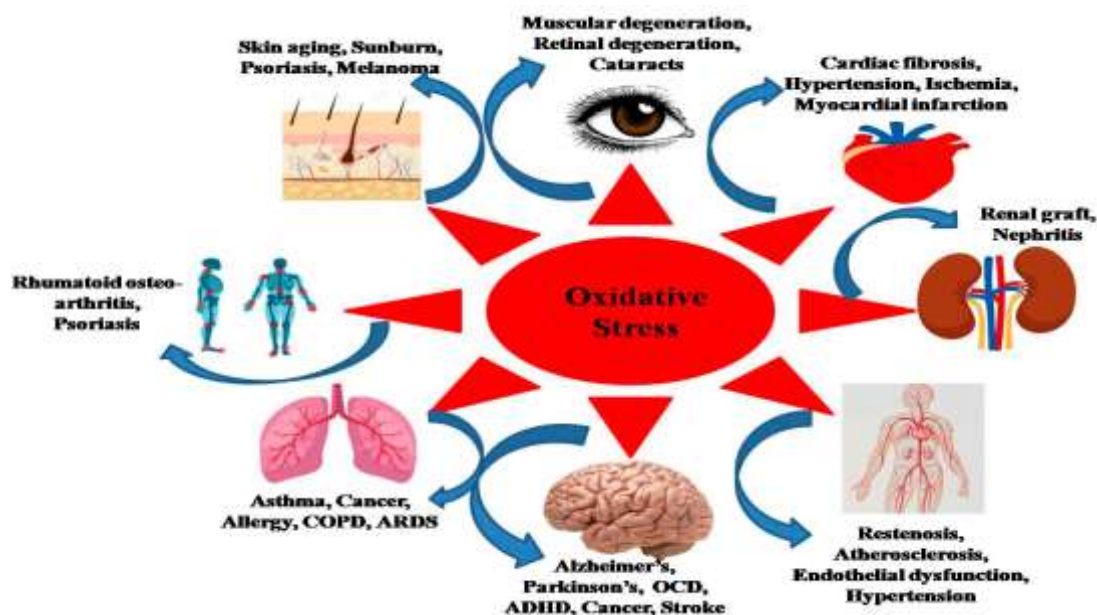
# I. Oxidative stress

## I.1. Introduction

Oxidative stress can be defined as a change in the balance between the production of reactive oxygen species (ROS) in the body that suppresses its ability to detoxify reactive intermediates, or to repair the damage to organs and cellular systems that can be caused by reactive oxygen species. The intracellular redox homeostasis (dynamic balance between intracellular redox species) is closely related to the antioxidant peptide glutathione; the intracellular levels of this compound are highly regulated by enzymes to maintain a reductive environment. Some of these ROS are constantly produced at low levels as by-products of normal metabolic reactions, and are controlled by enzymes that maintain redox balance within cells. When the redox balance is disturbed, these intermediate-reactive species may react with transition metals or other components of the redox cycle to produce highly reactive oxygen species that can cause extensive damage to lipids, proteins, DNA, and cellular organelles in cell membranes (Officer.,2007).



**Figure 1.** Basic concepts and components of oxidative stress. (a) oxidative stress is distinguished as either oxidative eustress or oxidative distress, depending on the balance between oxidants and antioxidants; (b) sources of oxidants and antioxidants; (c) effect of oxidative damage on biomolecules of DNA, proteins, and lipids. Abbreviation: DNA, deoxyribonucleic acid (Han et al., 2021).



**Figure 2.** Effects of oxidative stress on the human body (Kumar et al., 2020).

**Table 1.** Major Active Oxygen Species (Betteridge., 2000).

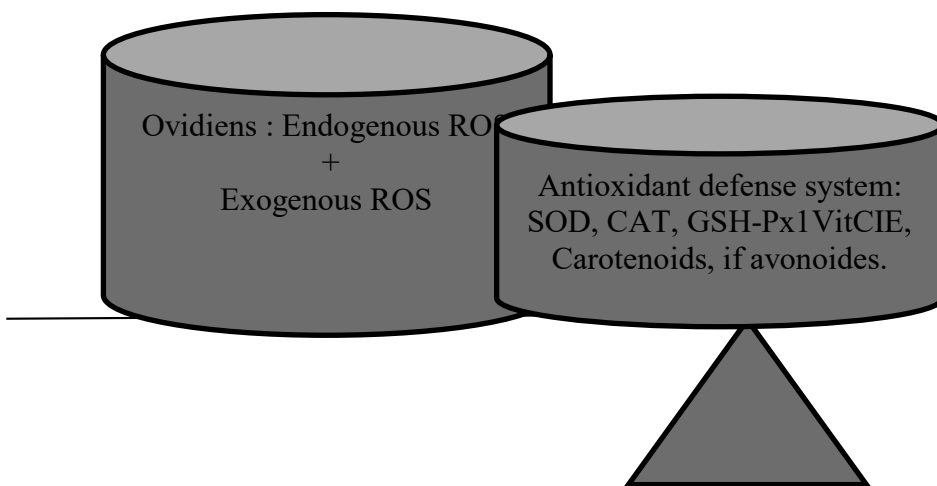
$O_2^-$	Superoxide radical
$H_2O_2$	Hydrogen peroxide
$HO^\bullet$	Hydroxyl radical
$^1O_2$	Singlet oxygen
$HOO^\bullet$	Hydroperoxyl radical
$LOOH$	Alkylhydroperoxide
$LOO^\bullet$	Alkylperoxyl radical
$LO^\bullet$	Alkoxy radical
$ClO^-$	Hypochlorite ion
$(Fe^{4+})O$	Ferryl ion
$(Fe^{5+})O$	Perferryl ion
$NO^-$	Nitric oxide

### I.2. Free Radicals

Free radicals are a single electron molecule that forms when a chemical bond is broken. 95 to 98% of the oxygen consumed is used in oxidative energy processes, and the remaining 2 to 5% occurs spontaneously in radical processes (Qu et al., 2010). These free radicals are usually unstable and highly reactive due to the unpaired electrons and tend to form pairs with other

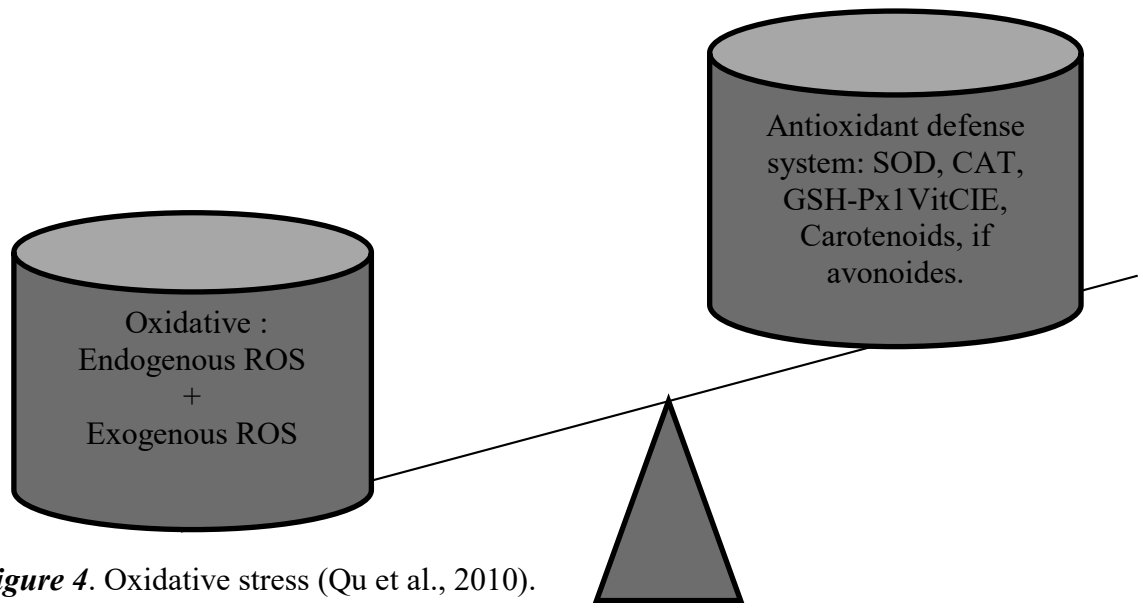
electrons. Oxygen ( $O_2$ ) has four electrons which is reduced and thus received in vivo and through this process activates oxygen metabolites formed by excitation of simple electrons to add energy or react with transition elements (Betteridge., 2000).

The production of small amounts of free radicals is normal. On the other hand, it is completely controlled by the body's defence system made up of many antioxidants. The endogenous compounds and enzymes such as superoxide dismutase (SOD), catalase (CAT), glutathione peroxidase (GPx) and small molecules such as vitamin E and vitamin C when the situation is under control and stable, we can say that the proportion of pro-oxidants in equilibrium, As shown in the figure 3 (Qu et al., 2010).

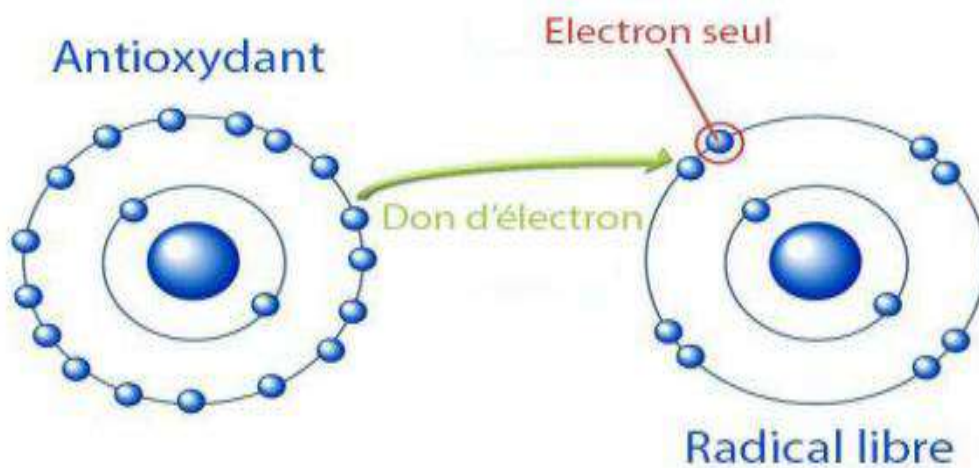


**Figure 3.** Antioxidant/pro-oxidant balance, SOD: superoxide dismutase, CAT: catalase, GSH-Px: glutathione peroxidase, Vit. C, E: vitamin C and E (Qu et al., 2010).

The situation becomes satisfactory when a system is no longer able to control production and formation of reactive oxygen species (ROS) causing disease and side effects and we are then in a state of preference for SO, resulting in an imbalance between endogenous antioxidants and ROS formation (Qu et al., 2010).



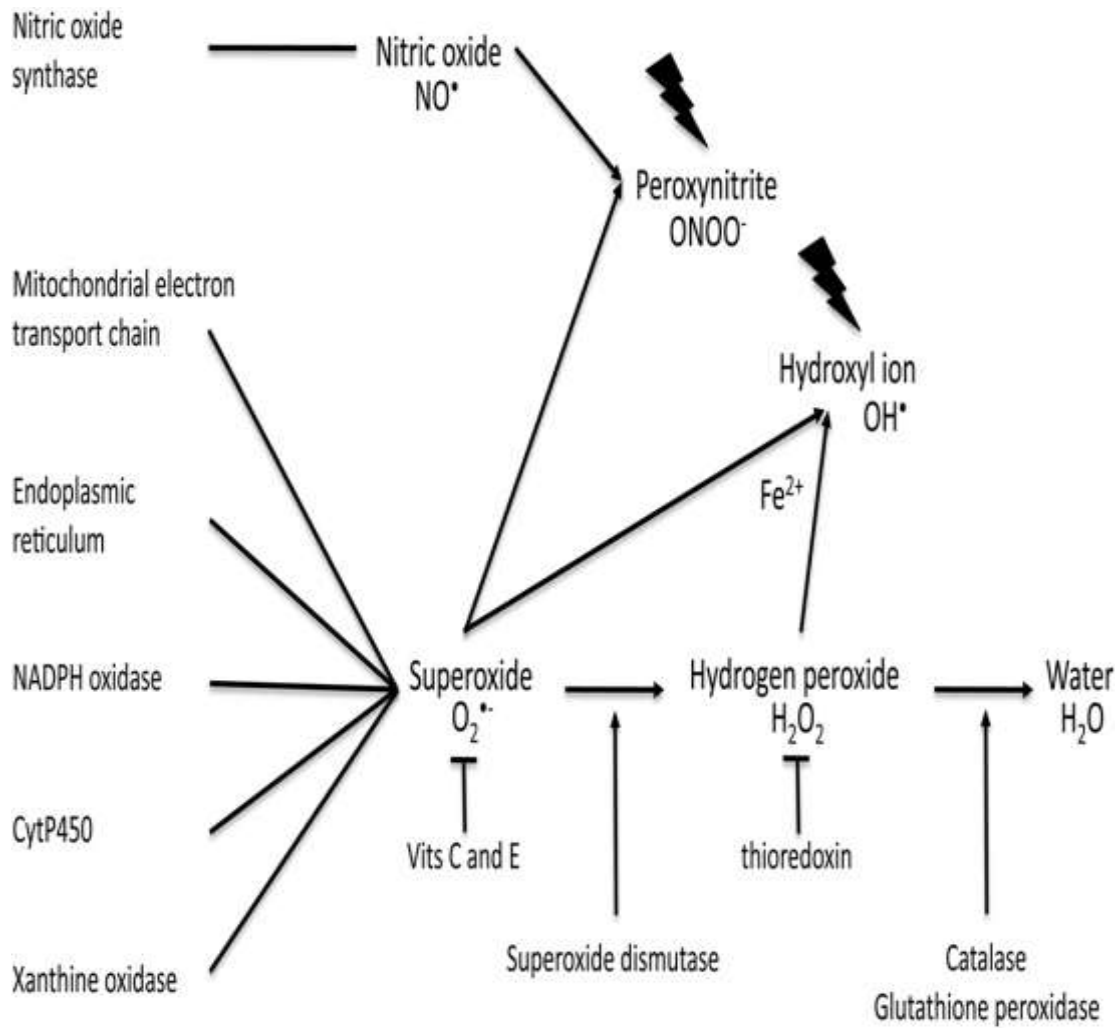
**Figure 4.** Oxidative stress (Qu et al., 2010).



**Figure 5.** Neutralization of a free radical by an antioxidant (Minist et al., 2021).

### I.3. Reactive species

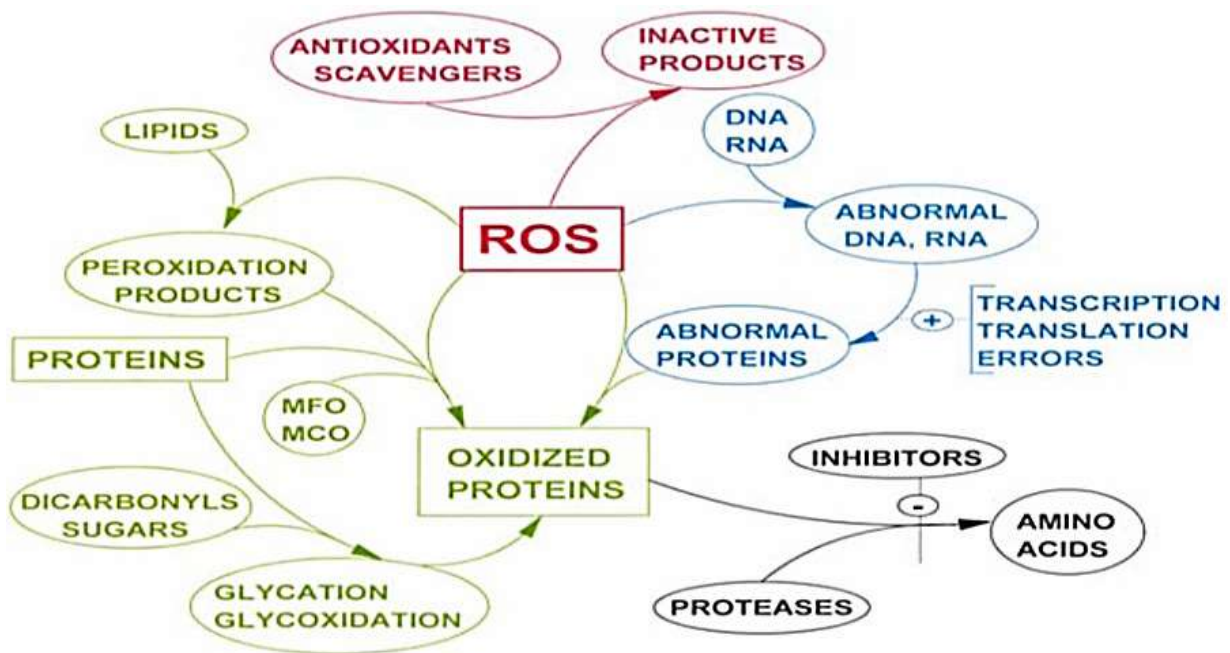
Reactive species and free radicals They are generated in the human body and some of them are made from “accidents of chemistry example: electrons leak directly into O<sub>2</sub> from intermediate electron carriers to generate a stable mitochondrial electron transport chain”(Halliwell.,2005).



**Figure 6.** Main Reactive Oxygen Species (Burton & Jauniaux., 2011).

### I.3.1. Reactive oxygen species (ROS)

The term "reactive oxygen species" is applied to both free radicals and all non-radical mediators and is used to define Reactive oxygen species containing oxygen (Burton & Jauniaux, 2011). Reactive oxygen species contain superoxide ( $O_2^{\bullet-}$ ), hydrogen peroxide ( $H_2O_2$ ), hydroxyl radicals ( $OH^{\bullet}$ ), monooxygenase ( $^1O_2$ ), peroxy radicals ( $LOO^{\bullet}$ ), Alkoxy radicals ( $LO^{\bullet}$ ), lipid Hydroperoxide ( $LOOH$ ), peroxynitrite ( $ONOO$ ), Hypochlorous acid ( $HOCl$ ), and ozone ( $O_3$ ) (Ahmed & Mohammed., 2020).

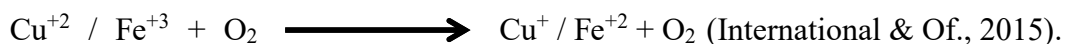


**Figure 7.** Effects of ROS (Ahmed & Mohammed., 2020).

### I.3.1.1. Types of ROS

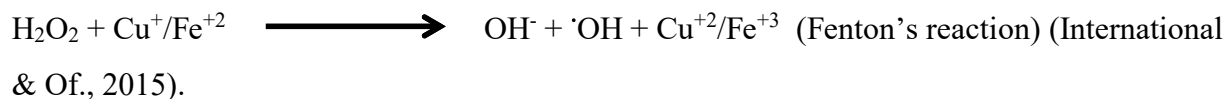
#### I.3.1.1.1. Superoxide

Superoxide is not very reactive, and is considered a reducing agent because it converts the form of iron ( $\text{Fe}^{+++}$ ) to form ( $\text{Fe}^{++}$ ). We can find superoxide in the place of its manufacture because it does not have the ability to enter and penetrate lipid membranes. The superoxide itself is an agent product, and it occurs particularly in the electron-rich aerobic environment of the inner membrane of mitochondria with the respiratory chain. Shaping and shaping from superoxide and hydrogen peroxide occurs at an endogenous level through flavin enzymes, that is, xanthine oxidase generally activates ischemic fusion (International & Of., 2015).



#### I.3.1.1.2. Hydroxyl radical ( $\cdot\text{OH}$ )

The hydroxyl radical is highly reactive and considered one of the most reactive types of oxygen, damaging or destroying biomolecules such as DNA, proteins, carbohydrates, and lipids. The hydroxyl radical of  $\text{H}_2\text{O}_2$  is formed through the Fenton reaction in which  $\text{H}_2\text{O}_2$  reacts with protein and other biomolecules containing transition metals ( $\text{Fe}^{+2}$  or  $\text{Cu}^{+}$ ) (International & Of, 2015).



### I.3.1.1.3. Hydrogen Peroxide (H<sub>2</sub>O<sub>2</sub>)

H<sub>2</sub>O<sub>2</sub> is a pale blue covalent liquid that mixes freely with water and acts as a light oxidizing agent and a light reducing agent. It reacts with proteins and other molecules including transition metals, but does not oxidize most biomolecules in an easy way. Our bodies manufacture H<sub>2</sub>O<sub>2</sub> to fight and kill pathogens, and it also stimulates our immune system to work properly. Neutrophils are white, blood cells Hydrogen peroxide production as a first line of defense against toxins, parasites bacteria, viruses and yeast (International & Of., 2015).

### I.3.1.1.4. Hypochlorite (HOCl)

When a reaction occurs between H<sub>2</sub>O<sub>2</sub> and chlorine, it leads to the formation of the most reactive ROS hypochlorite (International & Of, 2015).



**Table 2.** Sources of Reactive oxygen species in the body (Ahmed & Mohammed., 2020).

Sources	Description
<b>Endogenous</b>	NADPH oxidases Mitochondria Xanthine oxidoreductase Cytochrome P450 oxidases Nitric oxide synthase Peroxisomes
<b>Exogenous</b>	Physical agents Xenobiotic Biologic agents

### I.3.1.2. Sources of ROS

The main process of producing reactive oxygen species in aerobic exercise living organisms by reducing oxygen during electronic respiratory chain. ROS process occurs in different endogenous types, such as NADPH oxidases Mitochondria, external factors, including physical Factors, abiotic metabolism, and biological factor (Ahmed & Mohammed., 2020).

#### **I.3.1.3. Reactive nitrogen species (RNS)**

Reactive nitrogen species (RNS) are a group of defiant nitrogen molecules as well as free radicals, nitric oxide (NO) and nitrogen dioxide (NO<sub>2</sub>). The types of free radicals are nitric oxide radicals (NO) and nitrogen dioxide (NO<sub>2</sub>), non-radical elements are nitrite (NO<sub>2</sub>) and nitrate (NO<sub>3</sub>), and RNS is synthesized from its derivation by nitric oxide (NO) and superoxide anion (O<sub>2</sub>) (Tanaka & Vécsei., 2020).

#### **I.3.1.4. Reactive sulfur species (RSS)**

Reactive sulfur species (RSS) is a redox-active sulfur-based compound capable of oxidizing or reducing biomolecules, consisting of thiols (RSHs) and disulfides (RSSHs). RSS contains cysteine, methionine, GSH, trypanothione, mycothiol and Thiyl radicals (RS), highly reactive oxidants that form in the active regions of enzymes (Tanaka & Vécsei., 2020).

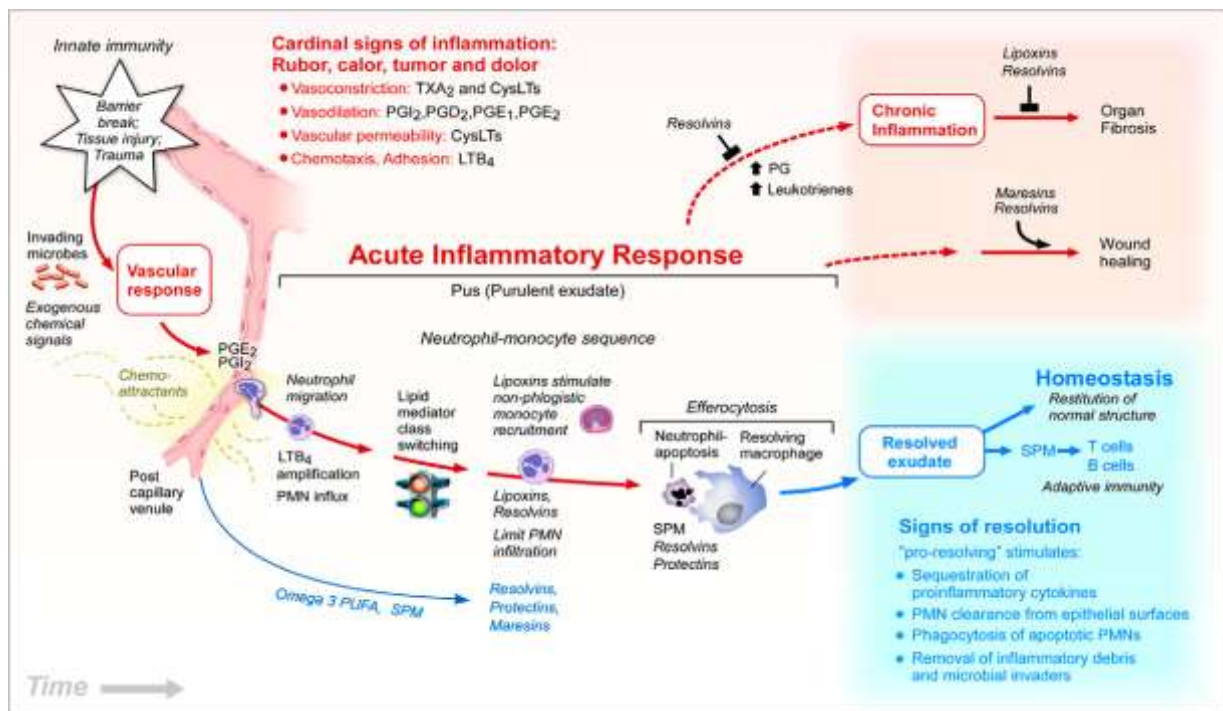
#### **I.3.1.5. Reactive Carbonyl Species**

Reactive carbonyl species (RCS) are formed through the metabolism of highly reactive molecules Aldehydes and electronically excited tricarbonyls (3 L = x \*), which are known for their harmful interactions with Nucleic acids, proteins and lipids, and RCS is a component involved in electrophilic Adaptive cell response signaling and protein modification and development after transcription. RCS classified into unsaturated aldehydes; Keto Aldehyde and D-Aldehydes (Tanaka & Vécsei., 2020).

### **I.4. Inflammation**



Inflammation is a local response (reaction) of living vascularized tissues to endogenous and exogenous stimuli. The term is derived from the Latin "inflammare" meaning to burn. Inflammation is fundamentally destined to localize and eliminate the causative agent and to limit tissue injury. Thus, inflammation is a physiologic (protective) response to injury. Inflammation is itself not to be considered as a disease but as a salutary operation consequent either to some violence or to some diseases” (Ansar & Ghosh.,2016).



**Figure 8.** Proresolving mediators in the inflammatory response (Serhan.,2017).

#### I.4.1. Causes of inflammation

physical agents - mechanical injuries, alteration in temperatures and pressure, radiation injurie, chemical agents- including the increasing lists of drugs and toxins (Andey., 2015), biological agents (infectious)- bacteria, viruses, fungi, parasites(Parag Jain, ravindra Pandey, 2015), immunologic disorders, hypersensitivity reactions, autoimmunity, immunodeficiency states etc... (Parag Jain, ravindra Pandey, 2015), genetic/metabolic disorders (gout, diabetes mellitus) etc...(Parag Jain, ravindra Pandey, 2015).

#### I.4.2. Types of inflammation

Inflammation is classified based on the duration of the injury and the histological manifestations of acute and chronic inflammation.

#### **I.4.2.1. Acute inflammation**

is an immediate and early response to a harmful agent and is of relatively short duration, lasting for minutes, several hours, or a few days.

It is characterized by the secretion of fluid and plasma proteins and the migration of predominantly neutrophilic leukocytes to the site of infection(Parag Jain, ravindra Pandey, 2015).

#### **The five main signs of acute inflammation are:**

1. Redness (friction) caused by dilation of small blood vessels within damaged tissue as occurs in cellulitis(Parag Jain, ravindra Pandey, 2015),
2. Heat (calories) caused by increased blood flow (congestion) due to the dilation of blood vessels in the area(Parag Jain, ravindra Pandey, 2015),
3. Swelling (tumor) caused by the accumulation of fluid in the extravascular space which in turn is due to increased vascular permeability(Parag Jain, ravindra Pandey, 2015),
4. Pain (pain), which is caused partly by the stretching and destruction of tissues by inflammatory edema and partly by pus under pressure in the abscess cavity. Some acute inflammatory chemicals, including bradykinin, prostaglandins, and serotonin, are also known to cause pain(Parag Jain, ravindra Pandey, 2015),
5. Loss of function: the inflamed area is suppressed by pain while severe swelling It may also physically immobilize tissues(Parag Jain, ravindra Pandey, 2015).

Acute inflammation is classified into early vascular response and late cellular responses(Parag Jain, ravindra Pandey, 2015).

#### **I.4.2.2. Chronic inflammation**

Chronic inflammation is a prolonged inflammatory process (weeks or months) where an active inflammation, tissue destruction and attempts to repair are proceeding simultaneously (Wakefield & Wales, 2014)

**Table 3 :** Markers correspond to both acute inflammation (Wu & Wu, 2007).

<b>Inflammation Reaction</b>	<b>Major Inflammatory Events</b>	<b>Associated Markers</b>
<b>Acute inflammation</b>	Production of proinflammatory cytokines.	TPN- $\alpha$ , IL-6
	Synthesis of acute phase reactants by Hepatocyte	CRP, SAA, fibrinogen
	Others	MCP-1, nitric oxide

**Table 4.** Markers correspond to chronic inflammation (Wu & Wu, 2007).

<b>Inflammation Reaction</b>	<b>Major Inflammatory Events</b>	<b>Associated Markers</b>
<b>Chronic inflammation</b>	Expression of adhesion molecules	VCAM-1, ICAM-1, E-selectin
	Microalbuminuria	Uma
	Released from activated leukocytes	MPO, ROS
	Nitrosative stress	3-Nitrotyrosine
	Oxidative stress	Urinary 8-OHdG
	Lipid peroxidation	Pospholipase A2, Urinary F2 isoprostane, COX-2

#### **I.4.3. Differentiation points between acute and chronic inflammations**

The different points between acute and chronic inflammations are:

**Table 5.** Differentiation points between acute and chronic inflammations include (Liu & Rabinovich, 2010) (Wakefield & Wales, 2014).

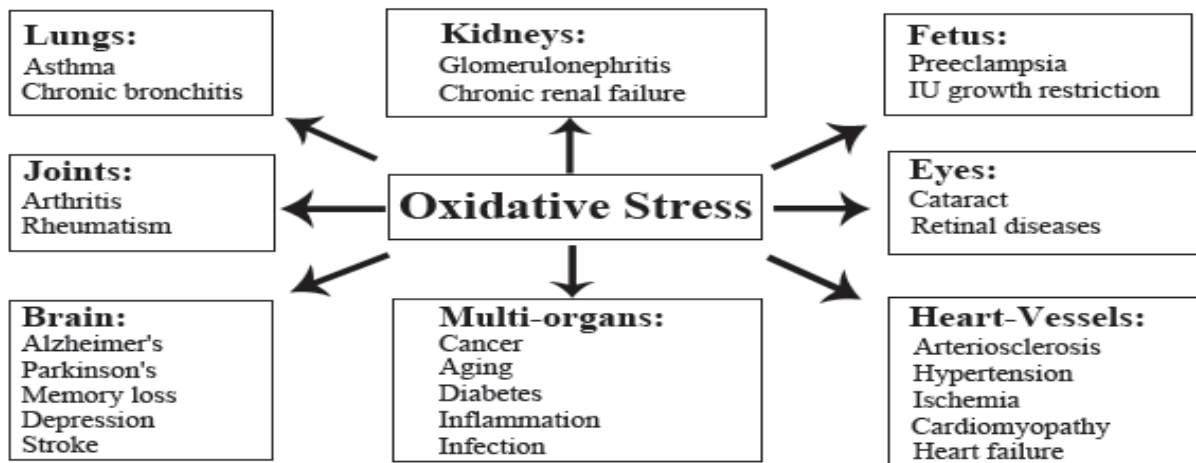
<b>Characteristics</b>	<b>Acute inflammation</b>	<b>Chronic inflammation</b>
<b>Duration</b>	Short	Relatively long
<b>Pattern</b>	Stereotyped	Varied
<b>Predominant cell</b>	Neutrophils	Plasma, cells Macrophages, Lymphocytes
<b>Tissue destruction</b>	Mild to moderate	Marked
<b>Fibrosis</b>	Absent	Present
<b>Inflammatory reaction</b>	Exudative	Productive

#### **I.4.4. The relationship between oxidative stress and inflammation**

The relationship between antioxidants and anti-inflammatories reactive oxygen species (ROS) and a weak antioxidant system promote the development of impaired inflammation and increased constriction of blood vessels, as it damages nerve cells and causes atrophy that leads to neuritis, cell destruction and death, nerve damage and memory loss. Glutathione tripeptide (GSH) is an intracellular thiol, an antioxidant that plays a role in regulation, off-label gene expression and associated free radical scavenging activities. Low level of GSH leads to increased level of ROS and this leads to unbalanced immune response and inflammation of all kinds as well as activation of defense mechanisms like nitrochophilus and phagocytosis.....etc (Hussain et al., 2016).

#### **I.5. Oxidative Stress and Disease**

Oxidative stress damages cells, tissues, and organ systems, resulting in serious diseases, including cancer, cardiovascular disease such as hypertension and atherosclerosis, and neurodegenerative diseases such as Parkinson's disease, Alzheimer's dementia, diabetes, ischemia/reperfusion injury, and rheumatoid arthritis. And even aging treatment (Officer., 2007).



**Figure 9.** Oxidative stress-induced diseases in humans (Pham-huy et al., 2008).

### **I.5.1. Inflammation**

Inflammatory responses are induced by noxious stimuli as reactive oxygen species (ROS) play a large role in the initiation of inflammatory processes by chemically heterogeneous free radicals (eg: superoxide) and non-radical radicals (eg: hydrogen peroxide) vital for cell growth. These mechanisms are rigidly regulated and when aberrant and inflammatory responses persist, they can lead to tissue damage and diseases such as arthritis, lung and kidney disease (Officer. 2007).

### **I.5.2. Cardiovascular disease**

Cardiovascular disease is closely related to ROS during development. Over-synthesis of ROS in vascular cells during reactions that involve enzymes such as NADPH oxidase and nitric oxide syntheses cause modification in LDL-k, and ROS have also been found to cause the development of cardiac hypertrophy, muscle apoptosis, ischemia and reperfusion injury, which ultimately leads to cardiac arrest (Officer.,2007).

### **I.5.3. Neurodegenerative disease**

Oxidative balance is a cause of many neurodegenerative diseases, such as Parkinson's disease, Alzheimer's dementia, as well as amyotrophic lateral sclerosis, and other neurodegenerative diseases such as bovine spongiform encephalopathy (BSE). When reactive oxygen species (ROS) are synthesized by dopamine oxidation in the central nervous system that correlates with age-related damage to dopamine neurons, Alzheimer's patients experience elevated levels of lipid peroxidation in the brain and increased levels of 4-hydroxynonenal: a

byproduct of lipid peroxidation in the brain. Cerebrospinal fluid. ROS have also been found to mediate beta-amyloid-induced neuronal damage (Officer.,2007).

### **I.5.1. Rheumatoid arthritis**

Induction or abnormal activation of oxidative stress-sensitive pathways leads to elevated reactive oxygen, resulting in the induction and activation of monocyte and lymphocyte migration into synovial rheumatoid arthritis and disease (Officer., 2007).

### **I.5.2. Ischemia/reperfusion injury**

Synthesis of reactive oxygen species leads to injury due to blood flow to tissues after ischemia has occurred. The restoration of circulation elicits a very significant inflammatory response and clearly stimulates the synthesis of reactive oxygen species thus reducing the occurrence of reperfusion and adherence of leukocytes to the endothelium (Officer.2007).

### **I.5.3. Diabetic**

Oxidative stress leads to the oxidation of glutathione in the blood through several mechanisms that activate interactions of glycoproteins with cell surface receptors in ROS synthesis resulting in hyperglycaemia, a process that can ultimately lead to the formation of atherosclerotic lesions (Officer.2007).

### **I.5.4. Aging**

Oxidative stress causes an increase in the production of reactive oxygen species, which leads to the phenomenon of aging. This is due to the occurrence of an oxidation process that indirectly leads to the formation of lipid peroxidation and oxidation of DNA (Officer.2007).

### **I.5.5. Cancer**

Reactive oxygen species play an important role in cancer because they facilitate mutation and potentiate tumor formation and growth. Growth-promoting mediators are modulated by redox-sensitive signaling cascades. Hydrogen peroxide and superoxide promote cell proliferation and proliferation, and the expression of growth-related genes. Malignant cells form themselves by ROS after generating genes associated with a mutated phenotype in addition to the primary oncogenes H-Ras or mox1. Oncogenic mutations in Ras proteins are found in approximately 30% of malignancies, and the products of transgenes induce cell proliferation and suppress apoptosis. Another mechanism also associated with cancer is that we find other oxidative stress states, resulting from a redox state shift in thiols/disulfides. We find glutathione in cells mainly

in the reduced form (GSH) and this is due to the fact that reduced glutathione is caused by oxidative stress. Cancer cells characteristically show reduced glucose clearance, and high levels of anaerobic glycolysis can stimulate lactate production, which reduces intracellular low levels of glutathione(Officer.2007).

***Chapter II***

***Secondary metabolites and  
their biological activities***



Plant secondary metabolites can be classified into four major classes: terpenoids, phenolic compounds, alkaloids and sulfur-containing compounds (Guerriero et al., 2018).

## II.1. Polyphenols

Polyphenols are secondary plant metabolism products that are widely used in pharmacology, medicine, and agriculture as biologically active compounds. All organs of the most diverse plants (fruits, seeds, roots, bark, wood, and leaves) contained phenolic compounds. The diversity of their chemical structure is determined by the breadth of their distribution in the plant kingdom. Simple phenols, hydroxycinnamic and hydroxybenzoic alcohols, aldehydes and acids, flavonoids, stilbenes, and lignans and their derivatives, tannins, and lignin are all examples of phenolic compounds found in plants (Chkhikvishvili and Ramazanov., 2000). Polyphenols are a well-known class of phenolic systems distinguished by the presence of at least two phenyl rings and one or more hydroxyl substituents (Han et al., 2007). Polyphenols can thus be divided into flavonoids and nonflavonoids or further subdivided based on the number of phenol units in their molecular structure, substituent groups, and/or the type of linkage between phenol units.

Polyphenols are produced by plants and are widely distributed in plant tissues, where they primarily exist as glycosides. Although flavonoids' basic structures are aglycones (the nonsugar fragment of the corresponding glycoside), flavonoids can be found as either glycosides or aglycones (Cardona et al., 2013).

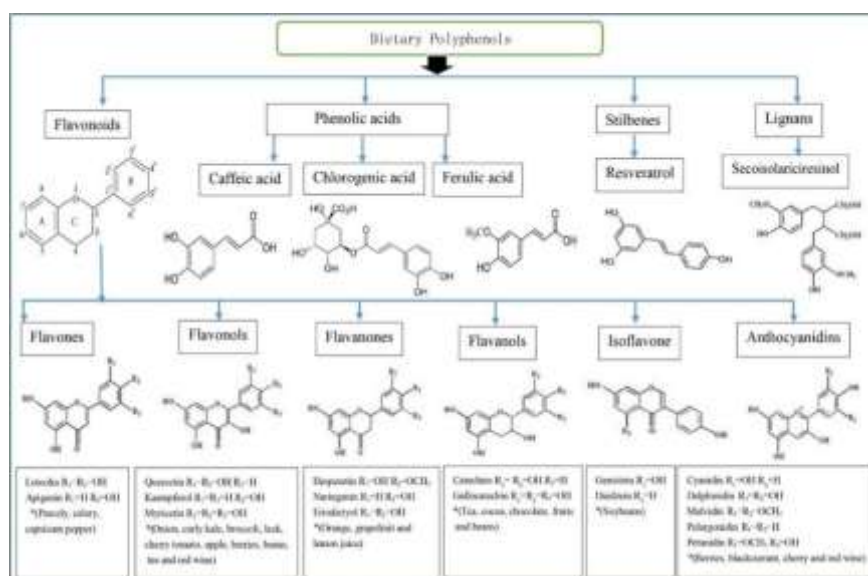

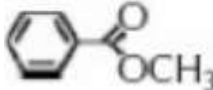
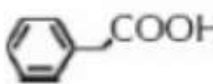
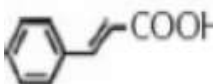
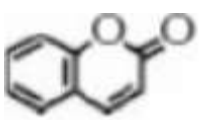
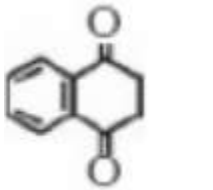
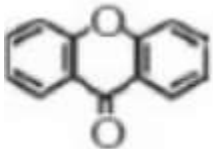
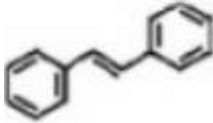
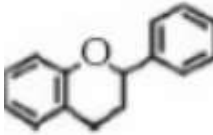


Figure 10. Classification of Polyphenols (Amira Namsi.,2019).

**Table 6.** Structure of polyphenol skeletons (Guerriero et al. 2018).

Number of carbons	Skeleton	Classification	Example	Structure
7	C6-C1	Phenol acids	Gallic acid	
8	C6-C2	Acetophenons	Gallacetophenone	
8	C6-C2	Phenylacetic acid	Hydroxyphenylacetic acid	
9	C6-C3	Acid hydroxycinnamic	Acid - coumaric	
9	C6-C3	Coumarins	Esculetin	
10	C6-C4	Naphthoquinones	Juglone	
13	C6-C1-C6	Xanthenes	Mangiferin	
14	C6-C2-C6	Stilbenes	Resveratrol	
15	C6-C3-C6	Flavonoids	Naringenin	

## **II.1.1. Non-flavonoid polyphenols**

### **II.1.1.1. Simple phenols**

Simple phenolic compounds are those that contain only one phenol unit (or a derivative of it). They are essentially substituted phenol compounds can be hydroxyphenyls or dihydroxybenzenes the general skeleton representation of simple phenolic compounds is C<sub>6</sub> (Cosme et al., 2020).

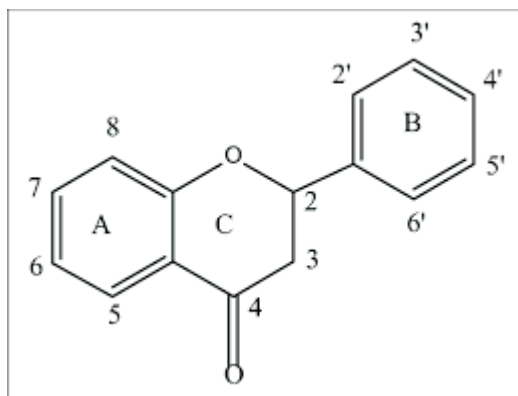
### **II.1.1.2. Phenolic acids**

Phenolic acids are aromatic secondary plant metabolites that are found all over the plant kingdom. Existing phenolic acid analytical methods arose from an interest in their biological roles as secondary metabolites, as well as their roles in food quality and organoleptic properties (Robbins., 2003). There are two types of phenolic acid: acid derivatives such as benzoic. acid and cinnamic acid derivatives (Grotewold 2006; Panche et al., 2016).

### **II.1.1.3. Flavonoids**

Flavonoids are a class of over 6,000 natural compounds found in nearly all vascular plants (Krogholm et al. 2010), They are pigments that cause the yellow, orange, and red coloration of various plant organs (Havsteen.,2002). Flavonoids are found in fruits and vegetables, especially in the genus Citrusm Red wine, tea, coffee, and beer all contain significant amounts. Flavonoids can also be found in a variety of medicinal plants. Flavonoids-containing herbal remedies have been (and continue to be) used in traditional medicine all over the world, all flavonoids are derived from the benzo-pyrone enzymatic reaction and can be classified according to their nature. substituents present on the molecule's cycles and the degree of saturation of the benzo-pyrone-squelette (Carlo et al. 1999).

Flavonoids are compounds in which the substitution by a benzene nucleus occurs in position 2, the term isoflavonoids refers to compounds that have a 3-position substitution (Glevitzky, M et al., 2019). The following are distinguished based on the nature of the heterocycle (i.e., the hydroxypyron or its dihydro-derivative): Flavones and flavonols, flavanones, flavonols, and dihydroflavanols (Panche et al. 2016).



**Figure 11.** Flavonoid structure (Michelle Lynn Wright et al., 2022).

**Table 7.** Classes of flavonoids (Krishna et al., 2001)

Classes	Chemical Structures	R3'	R4'	R5'	Exemples
Flavones		H	OH	H	Apigenin
		OH	OH	H	Luteolin
		OH	OCH3	H	Diosmetin
Flavonols		H	OH	H	Kaempferol
		OH	OH	H	Querctin
		OH	OH	OH	Myrecetin
Flavanols		OH	OH	H	Catechin
Flavanones		H	OH	H	Naringenin
		OH	OH	H	Eriodictyol

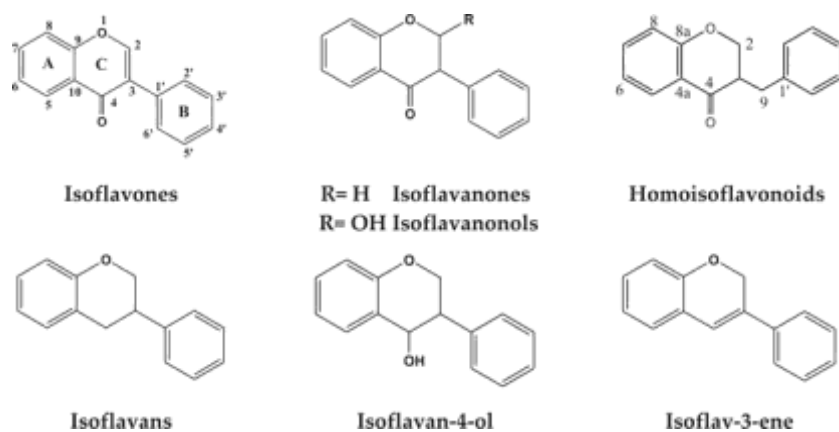
Anthocyanidines		H	OH	H	Pelargonidine
		OH	OH	H	Cyanidin
		OH	OH	OH	Delphinidin
Isoflavones		R5	R7	R4'	
		OH	OH	OH	Genistein
		H	O-Glu	OH	Daidzein

Flavones, Flavonols, Flavanones and Flavanonols are subgroups of flavonoids and are considered among most frequent in the kingdom plantae. Flavones and their 3-hydroxy derivatives flavanols, including glycosides, methoxides, and other acylated products on all three rings, constitute the largest subgroup of polyphenols. Quercetin and kaempferol, the two most common flavonol aglycones, have at least 279 and 347 different glycosidic combinations, respectively. In the last 15 years, the number of flavanones and their 3-hydroxy derivatives (flavanonols, also known as dihydro flavonols) identified has significantly increased.

Some flavanones, such as prenylated flavanones, furanoflavanones, pyranoflavanones, and benzylated flavanones, have distinct substitution patterns, resulting in a large number of substituted derivatives of this subgroup. Taxifolin, derived from citrus fruits, is a well-known flavanonol. (Panche et al., 2016).

#### II.1.1.4. The isoflavonoids

The isoflavonoids are a unique subclass of flavonoids. These substances possess a 3-phenylchroman skeleton derived biogenetically from 1,2-aryl, in a 2-phenylchroman precursor, migration occurs. Despite their limited availability. In terms of structural variation, isoflavonoids in the plant kingdom are remarkably diverse. This is due to the number and complexity of substituents on the basic 3-phenylchroman system, as well as the different oxidation levels and the presence of additional heterocyclic rings (Grotewold.,2006). Isoflavonoids are classified into four groups:



**Figure 12.** The skeletons of the isoflavonoids with three-ring structures (2022 ).

### II.1.1.5. Anthocyanins

Anthocyanins are flavonoid polyphenolic pigments that are responsible for many of the red-orange to blue-violet color found in plant organs such as fruits, flowers, and leaves. In nature, 700 structurally distinct anthocyanin derivatives of 27 aglycons known as «anthocyanidins» have been identified. Anthocyanins have recently piqued the interest of researchers due to their potential preventive and/or therapeutic effects on human health. Studies on the presence of anthocyanin in plasma and urine after dietary intake indicate a low bioavailability of about 1% (Czank et al.; Novotny et al., 2012).

Polyphenols in the form of polymers.

### II.1.1.6. Tannins

Tannins are nutritionally important plant secondary compounds that are complex phenolic organic molecules with molecular weights ranging from 500 to 3000 kDa (Cosme et al., 2020).

They are traditionally classified as hydrolysable tannins or condensed tannins (P. Frutos., 2004) Tannins are a heterogeneous group of phenolic compounds with high molecular weight that can form reversible and irreversible complexes with proteins (primarily), polysaccharides (cellulose, hemicellulose, pectin, etc.), alkaloids, nucleic acids, and minerals. (Philippe Lebreton 1982; N. Zimmer., 1996).

### II.1.2. Biological proprieties of Polyphenols

Phenolic compounds are found in both edible and non-edible plants and have been shown to have a variety of biological effects (B Halliwell.,1995) including antioxidant and antimicrobial agents, as oxidative enzyme inhibitors, as cell receptor modulators, anti-allergic,

anti-atherogenic, anti-inflammatory, hepatoprotective, antimicrobial, antiviral, antibacterial, anticarcinogenic, antithrombotic, cardioprotective, and vasodilatory(Middleton E Jr.,2000) Furthermore, as confirmed by several studies on the effect of dietary intake of polyphenol-rich foods, they may be beneficial with the prevention of cancer, cardiovascular and neurodegenerative diseases (Manach C 2005; Harri Vainio & Elisabete Weiderpass 2006; Pandey KB 2009).

### **II.1.3 Antioxidant Activity**

Natural polyphenols are the most abundant group of phytochemicals, and they are gaining popularity as potential agents for the prevention and treatment of oxidative stress-related diseases (Nelson Lugemwa et al., 2007) Natural polyphenols are secondary metabolites of plants that have been found in a variety of plant-based foods. Many *in vitro* experiments demonstrated that phenolic compounds were typically major contributors to plant antioxidant capacities. DPPH and NO scavenging (Li et al., 2014) revealed that rosmarinic acid, ferulic acid, caffeic acid, chlorogenic acid, vanillic acid, p-hydroxybenzoic acid, p-coumaric acid, protocatechuic acid, and other compounds contribute to the antioxidant potential of *Lycopus lucidus* and tea. Polyphenols may also act as antioxidants *in vivo* by influencing plasma, membranes, transcription factors, and enzyme activities. (Molay K Roy 1 2010; Rajeswari Ravindran., 2019)

Furthermore, the working mechanisms of natural polyphenolic antioxidants, namely H atom transfer, electron transfer, and metal ion chelation action, have been demonstrated in the literature.

### **II.1.4. Cardioprotection Activity**

Polyphenols provide cardioprotection through antioxidative, anticoagulation, and antiplatelet aggregation; fibrinolysis activity; activation of AMP-activated protein kinase, nitric oxide synthase, and sirtuin; inhibition of angiotensin-converting enzyme and phosphate diesterase; and improvement of endothelial cell function. Furthermore, flavonoids improve ventricular health, insulin resistance, and plasma lipid markers, have anti-inflammatory effects, and lower blood pressure to improve overall vascular health, as well as block cholesterol oxidation to lower LDL levels reduce atherosclerosis, and finally reduce the risk of cardiovascular disease. A similar effect is seen following resveratrol treatment (Otr, Eba et al., 2020).

### **II.1.5. Anticancer Activity**

Polyphenols may play an important role in the prevention of cancer in the mouth, stomach, duodenum, colon, liver, lung, mammary gland, and skin. Many polyphenols have been studied, including proanthocyanidins, flavonoids, resveratrol, tannins, epigallocatechin-3-gallate, gallic acid, and anthocyanin. (Johnson' et al., 1994).

Polyphenols have been found to affect several cancer prevention mechanisms, including oxidation prevention, xenobiotic detoxification, apoptosis induction, estrogenic/anti-estrogenic activity, stimulating effects on immune system function, anti-inflammatory properties, and their effects on the cellular signalling system. Among these are effects on nuclear factors such as NF-B and activator protein 1 (AP-1), which play critical roles in cellular signalling cascades by regulating DNA transcription, gene expression in response to various stimuli, cell proliferation, and survival. Despite the fact that their mechanisms of action were unknown, all of them demonstrated protective effects in some models. (Johnson' et al. 1994; Niedzwiecki et al., 2016).

#### **II.1.6. Anti-Inflammation Activity**

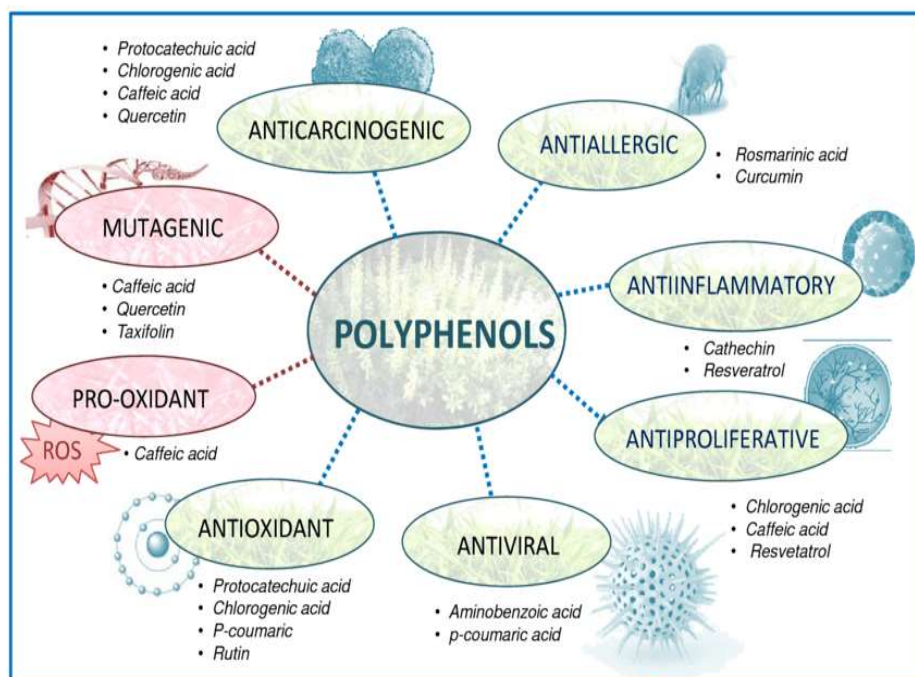
Several studies support polyphenols' immune modulation effect: some polyphenols influence immune cell populations, modulate cytokine production, and promote the expression of pro-inflammatory genes. (Santangelo C., 2007)

Polyphenols have anti-inflammatory and antibiotic properties, as well as the potential to activate the transcription factor Nrf2. Nrf2 is essential for cellular defence against oxidative stress and inflammation (Pandey K.B et al., 2019). Environmental factors have a significant impact on the dietary polyphenol content. Polyphenols such as quercetin, rutin, morin, hesperetin, and hesperidin have been shown to have anti-inflammatory properties in animal models of acute and chronic inflammation. (Santangelo C 2007; Hussain et al., 2016)

#### **II.1.7. Anti-Ageing Effect**

The primary cause of ageing is oxidative stress, which occurs during normal metabolism. Polyphenols have antiaging activity because of their significant antioxidant activity; they may be useful in reversing the course of neuronal and behavioral ageing. The researchers looked at how grape seed extract affected antioxidant status in different parts of the central nervous system of young and old rats (Chies et al. 2013)





**Figure 13.** Polyphenols and their biological properties (Ondrej Zitka et al. 2022)

## II.2. Terpenes

Terpenes of various sizes and compositions can be found in all classes of organisms. Terpenes appear to have evolved quite independently within small groups of organisms, resulting in a wide variety of structures capable of specific interactions with a variety of biological receptors. Many of the compounds used as research leads for pharmaceutical, agricultural, and other commercial applications come from plants with distinct terpenes (Ludwiczuk, A et al., 2017). Terpenes are classified structurally and functionally. They are isoprene is a compound formed by the combination of several 5-carbon-base (C5) units.

Monoterpenes (C10) and diterpenes (C20) are the most common terpenes found in medicinal plants. Diterpenes the monoterpenes are formed by the union of two isoprene units.

They are the most representative molecules constituting 90 percent of the essential oils and allow a great variety of structures. Oxygenated compounds derived from these hydrocarbons include alcohols, aldehydes, esters, phenols, and oxides. It is estimated that there are more than 1000 monoterpene structures (F Bakkali., 2008).

### **II.3. Alkaloids**

Pelletier defines an alkaloid as "a cyclic organic compound containing nitrogen in a negative oxidation state that has a restricted distribution among living organisms". The basic properties of most alkaloids are linked by a heterocyclic tertiary nitrogen. However, alkaloids have been discovered in microorganisms, marine organisms like algae, dinoflagellates, and puffer fish, and terrestrial animals like insects, salamanders, and toads. The two largest groups of alkaloids are the indole alkaloids and isoquinoline alkaloids. Tropane alkaloids, steroidal alkaloids, and pyridine and pyrrolizidine alkaloids are other important groups.

This can be explained in part by the structural similarities between dopamine, acetylcholine, noradrenaline, and serotonin. The existence Because alkaloids are water soluble in acidic conditions and lipid soluble in neutral and basic conditions, they have unique properties for medicinal use because they can be transported in protonated form and passthrough membranes in neutral form. Most synthetic medicines, in fact, contain one or more tertiary nitrogens (Zirintunda et al., 2022). Although plants were the first known source of alkaloid compounds, fungi, bacteria, insects, and animals now produce a diverse range of alkaloids as well. As a result, the definition of alkaloids has been expanded to state that "alkaloids are nitrogen-containing organic substances of natural origin with varying degrees of basic character. (Stella Omokhefe Bruce., 2021).

### **II.4. Molecules identified from *Hyacinthoideae* and *Artemisia* genres and their biological activity**

#### **II.4.1. From the genus *Hyacinthoideae***

The Hyacinthoideae have been thoroughly studied, with the bulbs receiving the most attention and the flowers receiving the most scrutiny. Homoisofavanones and spirocyclic nortriterpenoids are found throughout the subfamily, as are chalcones, xanthonenes, flavonoids, alkaloids, stilbenoids, chromones, norlignans, and benzopyranones (Mulholland., 2013)

These molecules have a variety of biological properties, including cytotoxic, antitumor, and cardiogenic activities, as well as anti-inflammatory, anti-cancer, anti-viral, and other beneficial properties. (Kolodziejczyk and Stochmal; Mottaghpisheh and Stuppner 2021).

#### **II.4.2 From the genus *Artemisia***

*Artemisia* species have a wide range of biological activities, including antimalarial, antitumor, and anti-inflammatory properties. (Abate et al., 2021) because of the significant potential of bioactive compounds such as phenolic compounds, terpenoids, and alkaloids, coumarins, acetylenes, sterols, caffeoylquinic acids, and the presence of anthocyanins and artemisinin. (Krishna et al., 2008)

***Chapter III***  
***Generalities about***  
***Hyacinthoides lingulata and***  
***Artemisia Judaica***

### III.1. Studied Plants

The decision to use these plants as the focus of a biological study was driven by: The dearth of prior studies on the biological activities and structural characterization of *S. lingulata* and *A. Judaica* renders our study original. the extensive use of *Artemisia* and *Scilla* species in traditional medicine to treat a variety of diseases. We were motivated by scholarly curiosity to explore the indigenous *S. lingulata* and *Artemisia Judaica* in search of significant findings after previous phytochemical studies revealed the presence of numerous secondary metabolites with strong biological effects.

#### III.1.1. *Artemisia judaica*

The Asteraceae family contains over 500 species, including the following: *Artemisia judaica* is an aromatic perennial herbaceous plant used in folk medicine in Algeria. (T. Dob and C. Chelghoum.,2006)

It is a plant with a such a hairy receptacle, flower heads arranged in long terminal clusters, and simple 1-2 cm long leaves. Florets inserted very obliquely on the achene, very hairy at the top, at least on hermaphroditic species (P. Quezel and S. Santa.,1963)

This plant is found in several regions, including mountains from 1200 to 1800 metres above sea level, the Algerian southern desert (from the desert to the coast), Egypt (desert and coast), and the Middle East (Sinai Peninsula, Saudi Arabia and Jordan) (T. Dob and C. Chelghoum., 2006) (M. Gast., 1989).

**Table 8.** Classification of *Artemisia Judaica* (M. Gharib., 2009)

Superdivision	Spermatophyta
Division	Magnoliophyta
Class	Magnoliopsida
Subclass	Asteridae
Order	Asterales
Family	Asteraceae
Genus	<i>Artemisia</i>
Species	<i>Artemisia Judaica</i>



**Figure 14.** *Artemisa Judaica* african plant database, (2022)

### **III.1.2. Chemical composition**

Several surveys have shown that this plant is rich in flavonoids and sesquiterpene lactones, also rich in other compounds such as piperitone and cinnamate, judaicin, cirsimaritin, terpinen-4-ol and bornyl acetate and also the presence of monoterpene hydrocarbons (santolina triene, artemisia ketone, and artemisia alcohol), also the presence of  $\alpha$ -phellandrene which is the most abundant. This plant also contains toxic compounds but in small quantities such as  $\beta$ -thujone and camphor. (T. Dob and C. Chelghoum., 2006) (C. Z. Liu et al., 2003) (N. Benmansour et al., 2016)

### **III.1.3. Biological and therapeutic activity**

It has been shown that the compound judaicin of this plant has a cardiotoxic effect, and that cirsimaritin, has an antiviral and antibacterial and antifungal action (inhibit the activity of *Staphylococcus aureus*, *Candida albicans* and *Rodotorula rubra*, *Callosobruchus maculatus*) as well as an inhibitory effect on the activity of several mammalian enzymes (C. Z. Liu et al., 2003) (S. M. Nofal et al., 2009) (S. A. M. Abdelgaleil et al., 2008) Its essential oil has insecticidal, anti-drug and antifungal (N. Benmansour et al., 2016) ,properties it also contains

santonin which has vermifuge effects . It has acaricidal, allelopathic and antioxidant activity (M. Gast., 1989)

The medicinal effects of this plant are: ophthalmic treatment, improvement of immunity, reduction of the risk of atherosclerosis, cancer, arthritis and gastrointestinal diseases (P. Janačković et al., 2015) anthelmintic, anti-inflammatory, analgesic and antipyretic (S. M. Nofal et al., 2009).

### ***III.2. Hyacinthoides lingulata***

*Hyacinthoides* Heist. ex Fabr. is a small enigmatic genus of approximately ten species in the subfamily Hyacinthoideae of Hyacinthaceae a own circle of relatives dealt with currently as a part of the Asparagaceae Phylogenetic research determined robust help for the monophyly of the genus and certainly separated it from *Scilla* L. The difference of *Hyacinthoides* and different genera belonging to the Hyacinthoideae had been in the beginning mounted the usage of morphological features. Characters used to split *Hyacinthoides* from different genera of the subfamily consist of bulb morphology and floral bract (bracteole) number Members of the genus are generally recognized as ‘bluebells. (Grundmann et al., 2010)

The floral bracts are absent or rudimentary. The leaves generally do not always appear at the same time as the flowers. The ovary has biovulated compartments. The anthers are purple-black. The floral bract is clearly marked. The leafy plant can flower sometimes, and the one with flowering stem is the only one to develop. The leaves are linear or filiform, 1 to 2 cm wide and 10 to 12 cm long. They are not hispidules. The plant is 5 to 30 cm long. The flowering stem is straight or bent to the side. The flowers are of variable colors (blue, lilac, pink or white). The seeds are dull black. They grow in forests, pastures, very common in the Mediterranean sub-atlas (Becal el Far) (Sadoudi Mahdi et al., 2016)

Quézel and Santa proposed a synthetic work, the New Flora of Algeria and southern desert regions. Of the 14 species of *Scilla* L. recognized in North Africa, seven, three of which are endemic, are found in Algeria. These species are widespread from the Tell to the Hauts-Plateaux and the northern slope of the Saharan Atlas. They thrive in open and more or less cool biotopes such as meadows, pastures and forest clearings. (Quezel, P. and Santa S., 1962)



**Figure 15.** *Hyacinthoides ligulata* pacificbulbsociety (2022)

**Table 9.** Botanical classification of *Hyacinthoides ligulata* (Cronquist et al., 1981)

Reign	Plantae
Class	Liliopsida
Subclass	Liliidae
Ordre	Asparagales
Family	Hyacinthaceae
Genre	Hyacinthoides



Species	<i>Hyacinthoides lingulata</i>
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### III.3. Biological and therapeutic activity of *Hyacinthoides lingulata*

The genus *Hyacinthoides* (Asparagaceae) is considered a valuable source of medicinal plants in africa used to treat female infertility and constipation (Breyer-Brandwijk and Watt.,1962), and promoting blood circulation, as an anti-inflammatory agent and as an analgesic (Nishida et al.,2008). *Hyacinthoides lingulata* (Poir) Rothm (Syn: *Scilla lingulata* Poir). The plant was originally endemic to northern regions of africa In Algeria, it is called 'Becal el Far', and is used to treat menopause (Chermat and Gharzouli.,2015).

Many biological studies revealed the glycosidase inhibitory (Watson et al., 1997), anti-inflammatory (Du Toit et al. 2011), anti-cancer (Ghoran et al., 2016) and antioxidant (Nishida et al., 2013).

*Chapter IV*

*Materials and Methods*

## IV.1. Materials

### IV.1.1. Plant material and crude extract preparation

The aerial parts of *A. Judaica* were collected in the region of illizi, Djanet and *H. lingulata* from Constantantine A voucher specimen has been deposited in the herbarium of the Research Unit: VARENBIOMOL. Air-dried leaves and flowers were extracted with the UAE procedure. Briefly the plant material (15 g) was extracted with methanol (250 mL) using a sonicator Bandelin Sonorex RK510H (Germany) at 35 KHz and P180/320W for 2h at room temperature. The solvent was evaporated in vacuo to leave 3.5 g of residue. The two extracts were used in the assessment of the antioxidant and the anti-enzymatic properties.

## IV.2. Methods

### IV.2.1. Maceration

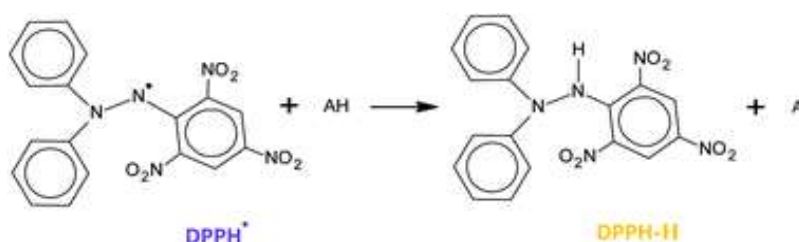
Maceration is a simple extraction method with the drawbacks of a long extraction time and low extraction efficiency. It has the potential to be used for the extraction of thermolabile components. It is a liquid-solid (solvent-vegetable powder) extraction method similar to infusion that is carried out at room temperature; it is commonly used for the extraction of heat-sensitive compounds. (Ćujić N, Šavikin., 2016).

### IV.2.2. Antioxidant Activity

Six different methods were used to assess the antioxidant capacity of the tested extracts:

### IV.2.3. DPPH Radical Scavenging Activity Assay

Marsden Blois introduced the DPPH method nearly 50 years ago; it is based on the use of the stable free radical diphenylpicrylhydrazyl (DPPH), which has a violet color and absorbs at 517nm. When a DPPH solution is mixed with a substance that can donate a hydrogen atom, the violet color is lost due to the reduced form of absorption. When this electron is paired off. (BLOIS 1958).



**Figure 16.** Application of free radical diphenylpicrylhydrazyl (DPPH) to estimate the antioxidant

Using a volume of 40  $\mu\text{l}$  of each extract at various concentrations was added to a microplate with 160  $\mu\text{l}$  of the freshly prepared DPPH ethanolic solution. At the same time, 40  $\mu\text{l}$  of ethanol and 160  $\mu\text{l}$  of DPPH ethanolic solution are combined to make the negative control.

After incubation in the dark and at  $-20^{\circ}\text{C}$  temperature, the absorbances are measured at 517 nm using a spectrophotometer against a blank containing 1ml of ethanol. Antiradical activity can be calculated by calculating the percentage inhibition of free radicals (I). by applying the following formula:  $I \% = [(Ac - At) / Ac] \times 100$

Where: I %: percentage inhibition

(At): Absorption of the test. (Ac): Control absorbance

#### **IV.2.4. Reducing Power Assay**

The reducing power of the aqueous extract was determined using the (Oyaizu 1986) method, which is based on the reduction of ferric ions  $\text{Fe}^{3+}$  to ferrous ions  $\text{Fe}^{2+}$ . The extract's reducing power is expressed as the number of equivalents of ascorbic acid that produces a green-blue coloration. At 700 nm, the absorbance of the reaction medium is measured.

The antioxidant effect was calculated by using the following equation:

Antioxidant effect (%) = (Control absorbance – Sample absorbance) / (Control absorbance)  $\times$  100.

The extract (10  $\mu\text{L}$ ) was added to a microtube of microplate and mixed with 40  $\mu\text{L}$  of  $\text{KH}_2\text{PO}_4$  buffer (pH 6.6) and 50  $\mu\text{L}$  of 1% potassium ferricyanide  $\text{K}_3\text{Fe}(\text{CN})_6$  and being incubated at  $50^{\circ}\text{C}$  for 20 minutes, 50  $\mu\text{L}$  of trichloroacetic acid 10% is being added after that with 40  $\mu\text{L}$  of distilled water and 10  $\mu\text{L}$  of ferric chloride the absorbance of the solution was measured at 700 nm.

#### **IV.2.5. ABTS Radical Cation Scavenging Activity Assay**

The ABTS radical cation scavenging activity was prepared according to the method developed by Re et al. (Re 1999). It is based on the reduction of ferric iron ( $\text{Fe}^{3+}$ ) in the  $\text{K}_3\text{Fe}(\text{CN})_6$  complex to ferrous iron ( $\text{Fe}^{2+}$ ), which, in the existence of an antioxidant capable of yielding electrons, results in the formation of a complex. This ABTS<sup>+</sup> assay was performed using a colorimetric method at 734 nm, and the extent of decolorization was calculated as percentage inhibition of the ABTS<sup>+</sup> radical cation as a function of concentration and time and was calculated relative to the reactivity of Trolox, as a standard. In brief, ABTS solution 19.2 mg (7 mM) was reacted with potassium persulfate (2.45 mM) solution and kept for 12-16 h in

the dark to yield a blue-green colored solution containing ABTS radical cation. Before use in the assay, the resulting blue-green ABTS<sup>+</sup> • solution was adjusted with methanol for an initial absorbance of about  $0.700 \pm 0.02$  at 734 nm, with the temperature control set at 30°C.

Free radical scavenging activity was assessed by mixing 40 µl of the test sample with 160 µl of ABTS radical solution microcuvette. The decrease in absorbance was measured exactly 10 min after mixing the solution. The final absorbance was noted. The scavenging effect was calculated according to the following equation:

$$\text{Scavenging effect (\%)} = \frac{(\text{Control absorbance} - \text{Sample absorbance})}{(\text{Control absorbance})} \times 100.$$

The antioxidant capacity of the test sample was expressed as EC50, the concentration necessary for a 50% reduction of ABTS.

#### **IV.2.6. Phenanthroline Assay**

The phenanthroline assay was investigated using the approach of (Szydłowska-Czerniak 2008). A volume of 10 L of plant extracts or standards solutions at various concentrations (3.125-200 g/mL) was added to a volume of 50 L FeCl<sub>3</sub> (0.2 percent). The mixture was then treated with 30 L of phenanthroline (0.5%) and 110 L of methanol. After incubating the solution for 20 minutes at 30°C, the absorbance at 510 nm was measured. As standards, BHA and BHT were used, and the results were expressed as A0.50 values (g/mL). Assay for scavenging galvinoxyl radicals (GOR). We used the protocol previously described by Shi et al. to evaluate this activity (2001). The mixture contained 40 L of *C. tougourensis* extracts or standards solutions at concentrations ranging from 3.125 to 200 g/mL and Galvinoxyl radical (160 L) (0.1 mM). After incubating the samples for 120 minutes, the absorbance at 428 nm was measured. As antioxidant standards, BHA and BHT were used, and the results were expressed as IC50 (g/mL).

#### **IV.2.7. Cupric Reducing Antioxidant Capacity**

With minimal adjustments, the cupric-reducing antioxidant capacity (CUPRAC 2004) method was used to measure the extracts' cupric-reducing antioxidant capacity. 40  $\mu$ l of samples and 50  $\mu$ l of  $\text{CuCl}_2$  were put into each well of a 96-well microplate (10 mM). After that, each well was given 50  $\mu$ l of neocuproine solution (7.5 mM) and 60  $\mu$ l of  $\text{NH}_4\text{Ac}$  buffer (1 M, pH 7.0). The absorbance at 450 nm was measured against a reagent blank after 60 minutes of incubation. In comparison to the absorbance of BHA and BHT, which were employed as antioxidant standards, the results are 0.5 (g mL<sup>-1</sup>) equivalent to the concentration, indicating 50% absorbance intensity.

#### **IV.2.8. Silver nanoparticle (SNP) assay**

The silver nanoparticle assay was determined using Mustafa Özyürek's method (Gu et al., 2012). The  $\text{AgNO}_3$  (1.0 mM), 1 percent trisodium citrate (w/v),  $\text{CuCl}_2$  (10.0 mM), and  $\text{NH}_4\text{Ac}$  buffer (1.0 M, pH 7) solutions were all prepared in pure distilled water (Millipore Simpaki Synergy 185, USA), and the neocuproine (7.5 mM) solution in absolute ethanol. All antioxidant compounds were freshly prepared in EtOH at 1.0mM concentration before measurement, as was L-ascorbic acid in distilled water at 1.0mM concentration. The SNPs were created using a chemical reduction method with trisodium citrate as a surface stabilizing agent. In this method, 19.5 mL of 1.0 mM  $\text{AgNO}_3$  were heated to boiling for 10 minutes. 5 mL of 1% trisodium citrate was added to this solution drop by drop. Throughout the process, the solution was vigorously mixed. This assay was evaluated by combining 20  $\mu$ l of the test extract with 130  $\mu$ l of SNP solution and 50  $\mu$ l of  $\text{H}_2\text{O}$ , then incubating it at 25°C for 30 minutes and measuring the absorbances at 423.

#### **IV.3. Anticholinesterase activity**

The inhibitory activities of acetylcholinesterase (AChE) were measured using a slightly modified spectrophotometric method (Ellman et al., 1961). The enzymes AChE from horse serum were used, and the substrates of the reaction were acetylthiocholine iodide and butyrylthiocholine chloride. The activity was measured using 5,5-Dithiobis (2-nitrobenzoic) acid (DTNB). As a positive reference compound, galantamine was used. The results were presented as the IC<sub>50</sub> value (mg/mL), which corresponds to the concentration that shows 50% inhibition.

#### **IV.4. The $\alpha$ -amylase inhibition activity**

The iodine/potassium iodide (IKI) technique was used to modify the inhibitory activity of  $\alpha$ -amylase (Gokhan, Sarikurkcu, Aktumsek, Ceylan, & Ceylan, 2014). 25  $\mu$ l of the sample was added to 50  $\mu$ l of  $\alpha$ -amylase solution (1U produced in phosphate buffer with 6 mM NaCl) in a 96-well microplate and incubated at 37°C. After 10 minutes of preincubation, the presence of a starch solution (50  $\mu$ l, 0.1 percent) aided the reaction's commencement. The reaction mixture was incubated at 37°C for 10 minutes before being stopped by adding 25  $\mu$ l HCl (1 M), which was followed by higher in water extracts and lowered in the order SAE DE > SVR DE > SAE ME > SVR ME SAE DE > SVR DE > SAE ME > SVR ME 100  $\mu$ l of the iodine-potassium  
Calculation

*Chapter V*

*Results and discussion*



### V.1. Antioxidant activity of *A. judaica* and *H. lingulata*

The antioxidant activity is a very appreciable property of extracts and plant-derived compounds. Because of the variable responses exerted by a specific antioxidant in various testing systems, it is important to use diverse antioxidant methods to take in account the mechanism of action of each compound (Moukette et al., 2015). Six different methods were performed to investigate the in vitro antioxidant activity of *A. Judaica* and *H. lingulata* extracts and the results were reported as IC<sub>50</sub> and A<sub>0.5</sub> values in Table 10

DPPH and ABTS are synthetic free radicals widely used to evaluate the anti-radical capacity of plant extracts and pure compounds. In the presence of antioxidant molecules, DPPH or ABTS<sup>+</sup> can accept an electron or a hydrogen atom from the antioxidant compound and will thus, be converted to a more stable molecule (Carmona-Jimenez et al., 2014). Spectrophotometrically, this reduction was observed by the color switch for DPPH radical (from purple to yellow) and decolorization for ABTS radical. According to Table 2, AG extracts showed a significant scavenging capacity of DPPH radical with IC<sub>50</sub> value of 85.39±2. whereas for ABTS scavenging activity, Both AJ and HL were found to be effective in reducing ABTS radical with values of 81.99±0.82 and 95,23±3,54 mg/ml, respectively. However, the tested extracts did not seem to exert a significant effect in the silver nano particles chelating activity or in the reducing.

The cupric reducing antioxidant capacity (CUPRAC) is based on the reduction of the neocuproine-copper complex, resulting in the formation of a chromogenic complex of Cu (II)-Nc, which absorbs at 450 nm (Saci et al., 2020). According to Table ..., the AG displayed a significant antioxidant activity with an IC<sub>50</sub> value of 72.08±15.21 whereas for HL the effect was weak (257.38±10.11): water Reducing property is one of the mechanisms exerted by antioxidant compounds, which can refer to the capacity of these compounds to regenerate another compound already oxidized by free radicals or to stop the free radical chain reaction (Jayaprakasha et al., 2001).

The analysis of metal iron-reduction assessed by phenanthrolines showed that the lowest value of A<sub>0.5</sub> was obtained with AG extract (33.91±12.60), while HL extract showed a higher

$A_{0.5}$  value ( $92.32 \pm 49.07$ ). These values and despite of being high, remain significantly different from the tested standards (BHA, BHT, Trolox and Vitamin C).

**Table 10.** the antioxidant activity of *A. judaica* and *H. lingulata*

SAMPLES	$IC_{50}$ ABTS ( $\mu\text{g/mL} \pm \text{SD}$ )	$IC_{50}$ DPPH ( $\mu\text{g/mL} \pm \text{SD}$ )	$A_{0.5}$ CUPRAC ( $\mu\text{g/mL} \pm \text{SD}$ )	$A_{0.5}$ reducing power ( $\mu\text{g/mL} \pm \text{SD}$ )	$A_{0.5}$ Snp ( $\mu\text{g/mL} \pm \text{SD}$ )	$A_{0.5}$ Phen ( $\mu\text{g/mL} \pm \text{SD}$ )
AJ	$81.99 \pm 0.82$	$85.39 \pm 2.02$	$72.08 \pm 15.21$	>200	>200	$33.91 \pm 12.60$
HL	$95.23 \pm 3.54$	>800	$257.38 \pm 10.11$	>200	>200	$92.32 \pm 49.07$
BHA	$1.81 \pm 0.10$	$6.14 \pm 0.41$	$3.64 \pm 0.19$	$0.93 \pm 0.07$	$1.05 \pm 0.01$	$0.93 \pm 0.07$
BHT	$1.29 \pm 0.30$	$12.99 \pm 0.4$	$9.62 \pm 0.87$	$2.24 \pm 0.17$	$0.90 \pm 0.02$	$2.24 \pm 0.17$
Ascorbic acid	$3.04 \pm 0.05$	$4.39 \pm 0.01$	$8.31 \pm 0.15$	$3.08 \pm 0.02$	/	$3.08 \pm 0.02$
Trolox	$3.21 \pm 0.06$	$5.12 \pm 0.21$	$8.69 \pm 0.14$	$5.21 \pm 0.27$	/	$5.21 \pm 0.27$

## V.2. Cholinesterase inhibitory Activity

Table 11 shows the BChE inhibitory activity of the extracts, compared with that of galantamine used as a standard drug to treat mild Alzheimer's disease. No significant inhibitory activity against BChE was shown by both AJ and HL extracts ( $IC_{50} > 200$  for AJ and HL).

**Table 11.** Butyrylcholinesterase inhibitory activities of *A. judaica* and *H. lingulata* . BChE assay

Samples	$IC_{50}$ ( $\mu\text{g/mL}$ )
AJ	>200
HL	>200
Galanthamine	$6.27 \pm 1.15$

AJ : *A. judaica* extract

HL: *H. lingulata* extract

$IC_{50}$  : 50 percent inhibition concentration

## V.3. Antidiabetic inhibitory activity of extracts

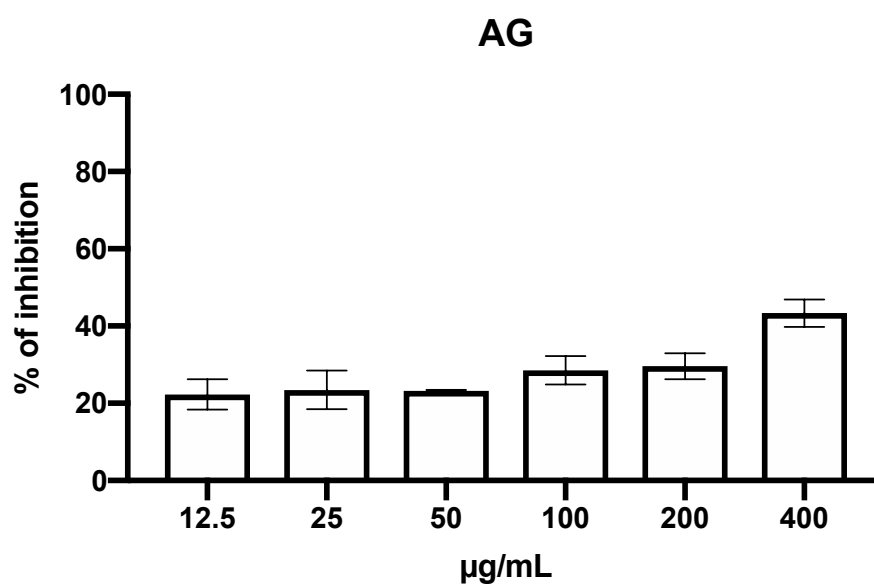
The investigation of the antidiabetic effect of *A. Judaica* and *H. lingulata* was carried out by evaluating the potential of the extracts to inhibit the enzyme  $\alpha$ -amylase. Table 12 represents the antidiabetic effect of the tested extracts reported in terms of  $IC_{50}$  values. Figures 17 And 18 the antidiabetic activity of AJ And HL extracts respectively expressed as percentages of inhibition. Compared to the control (Acarbose) no significant inhibitory effect was demonstrated by AJ extract ( $IC_{50} > 400 \mu\text{g/mL}$ ), however, a moderate effect was observed by HL extract at the high dose of  $200 \mu\text{g/mL}$  (%of inhibition=52.34).

**Table 12.**  $\alpha$  amylase inhibitory activities of *A. judaica* and *H.lingulata*.

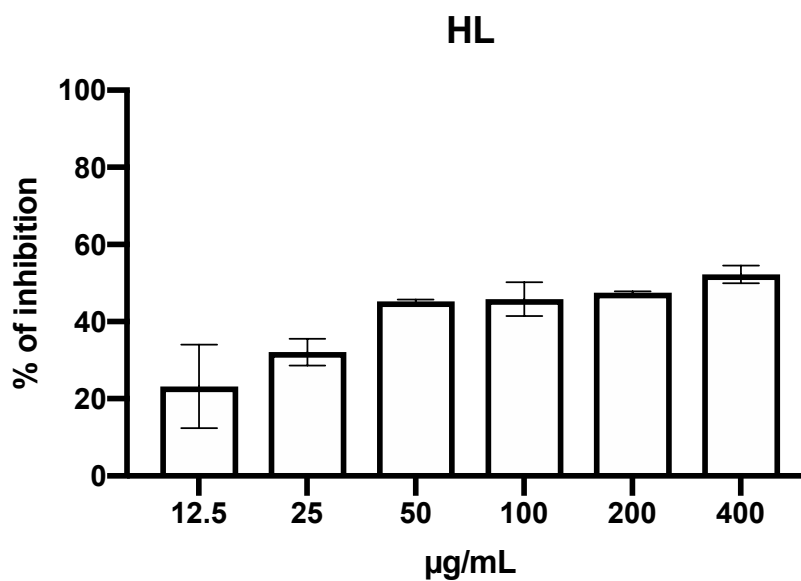
Samples	IC <sub>50</sub> ( $\mu\text{g/mL}$ )
AG	>400
HL	>400
Acarbose	257.43 $\pm$ 1.59

AJ : *A. judaica* extract

HL: *H. lingulata* extract



**Figure 17.**  $\alpha$  amylase inhibitory activities of *A. judaica*.



**Figure 18.**  $\alpha$  amylase inhibitory activities of *H.lingulata*

## V.4. Discussion

Plants and herbs' secondary metabolites are increasingly capturing a great interest as potent beneficial components of the human diet with a broad of traditional applications encompassing a variety of therapeutic properties related to their phytochemical composition (Dirar *et al.*, 2014).

Antioxidative effects of natural products could be considered as a first insight in detecting their ethnopharmacological relevance and potential. To this end, a certain antioxidant profile could be provided by comparing different chemical assays representative of alternative mechanisms (Zengin *et al.*, 2018).

Further more The  $\alpha$ -glucosidase inhibitors, such as acarbose and miglitol, impede certain enzymes responsible for the breakdown of carbohydrates in the small intestine. This class of hypoglycemic agents acts mostly by reducing the absorption rate of carbohydrates in the body. Also, acarbose reversibly inhibits both pancreatic  $\alpha$ -amylase and  $\alpha$ -glucosidase enzymes by binding to the carbohydrate-binding region and by interfering with their hydrolysis into monosaccharides, which leads to a slower absorption together with a reduction in postprandial blood sugar levels ( DeFronzo R.A.,1999 )(Inzucchi S.E.,2002).

Discovery of the new natural antidiabetic drugs could be great promise due to minimal efficacy and safety concerns of current antidiabetic drugs for the hundreds of millions of individuals which are currently seeking better management of diabetes (Mancha-Ramirez A.M., *et al.*,2016). In this relation, the investigation of phytochemicals responsible for antidiabetic effects has progressed in the last decades.

For this reason, the antioxidant ability of *A. Judaica* and *H. lingulata* was evaluated using five complementary *in vitro* tests: free-radical scavenging (DPPH, ABTS and OH.), reducing power, CUPRAC, phenanthroline, SNP assays. The results are summarized in Tables 10.

### V.4.1. Antioxidant and antiradical activity

The DPPH test was used to assess the antioxidant activity of our extracts. DPPH is a free, stable radical with an absorbance band at 517 nm that is used to assess the antioxidant activity of polyphenolic compounds. BHT is used as a standard in this test.

The DPPH radical's trapping capacity is determined by measuring the decrease in absorbance of a DPPH methanol solution at 517 nm.

Based on the results (Table. 10), we observed an increase in DPPH inhibition percentages as the concentrations of the extracts studied increased. The kinetics of this activity allow us to determine the percentages of maximum inhibition on the one hand, and the concentration that corresponds to 50% inhibition on the other.

the lowest IC<sub>50</sub> value corresponds to the highest extract efficiency. Observation of these results has allowed us to conclude that all the extracts have antiradicalar power.

The extracts EMet and EAq reach a maximum power of 85.39% These profiles are higher than that of the BHT which reaches a maximum power of 12.99%The statistical analyses for the DPPH test show that there is a significant difference between the two extracts ( $P > 800$ ) the free radical DPPH., the amount of polyphenols of the two extracts is different but the two extracts have almost the same effect, on the free radical DPPH. This can be explained by the fact that both extracts are rich in polyphenols. The results of cuprac assay 72.08% for *A. judaica* and 257.38 % *H. lingulata*, while BHT and BHA was low.

phenanthroline assay and reducing power shows to a significant activity accourdding to the results in table 10.

Because of their ability to produce hydrogen, antioxidant molecules such as ascorbic acid, tocopherol, flavonoids, and tannins have been shown to reduce and discolor DPPH (Bougandoura and Bendimerad 2012). The antioxidant activity in our extracts is most likely due to the polyphenols.

A wide range of opinions on this correlation has been discovered through bibliographic research. Some research has found a strong relationship between IC<sub>50</sub> and polyphenols and flavonoids (Athamena et al., 2010).

#### V.4.2. The Antidiabetic inhibitory activity of extracts

Herbal medicines provide an interesting, largely unexplored source for the development of potential new agents that can antagonize enzymes' activity. High level of carbohydrate intake is inevitably associated with an increased risk for obesity and type 2 diabetes.  $\alpha$ -Amylase and  $\alpha$ -glucosidase are two classes of the enzymes responsible for carbohydrate digestion resulting in postprandial high glucose level in the body (Habtemariam and Varghese, 2014).

The use of carbohydrate digestion enzyme inhibitors from natural resources could be a possible strategy to block dietary carbohydrate absorption and present an economical alternative to the oral synthetic hypoglycemic drugs with less adverse effects. Several studies on plant extracts of *A. Judaica* and *H. lingulata* have focused on the inhibition of  $\alpha$  glucosidase and  $\alpha$  amylase. which catalyze carbohydrate digestion into glucose In this study, we investigated the inhibitory effect of *C. fontanesii* extracts on  $\alpha$ -amylase to elucidate the potential of *A. Judaica* and *H. lingulata* as a natural agent in reducing postprandial hyperglycemia. *H. lingulata* extracts exhibited a mild inhibitory effect on  $\alpha$ -amylase, which suggested that *H. lingulata* extracts can be subjected could control diabetes by reducing postprandial hyperglycemia.

# *Conclusion*

## Conclusion

Much research has been conducted in order to identify natural molecules with biological activities derived from medicinal plants. Natural plant extracts containing a variety of phenolic compounds and flavonoids that are given. Because of their antioxidant activity, anti-diabetic, and anti-acetylcholinesterase properties, these natural phenolic molecules are highly sought after in herbal medicine.

Six different tests were used to assess antioxidant activity: the inhibition of the radical DPPH, the Reducing Power Assay, the ABTS Radical Cation Scavenging assay, the phenanthroline assay, the Cupric Reducing Antioxidant assay, and the silver nanoparticle (SNP) assay. The anticholinesterase activity of BChE was measured with 5,50-dithiobis(2-nitrobenzoic) acid (DTNB). The activity of  $\alpha$ -amylase inhibition and reductive power. For the first time, *A. Judaica* and *H. lingulata* extracts demonstrated that both EAq and emitting extracts have significant antioxidants due to their constituents (polyphenolic compounds).

Furthermore, they have interesting  $\alpha$ -amylase inhibitory activities with a mild effect. Therefore, *H. lingulata* and *A. judaica* may be interesting for further clinical and in vivo investigational studies for the treatment of diabetes mellitus, is needed to focus on the effects revealed. but also, by improving the antiradical defenses of the patients. And even molecular-scale studies are required to determine, on the one hand, the compounds that may be responsible for such effects and, on the other hand, the absolute mechanism by which these compounds perform their functions.

According to our findings, our studied plants possess a great importance in the medical, and food industries.



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## Abstract

Medicinal plants are understood as natural antioxidant sources, and thus play an important role in resolving health issues and treating diseases.

*Hyacinthoides lingulata* and *Artemisia Judaica*, are two medicinal plants that have long been known for their medicinal properties. We tested antioxidant activity, cholinesterase inhibitory activity, and anti-diabetic activity using extracts in this study.

Maceration is used to obtain the extracts. Six methods were used to test the antioxidant activity of the extracts : DPPH free radical scavenging assay, the Reducing Power Assay, ABTS Radical Cation Scavenging assay, phenanthroline assay, Cupric Reducing Antioxidant assay, and the silver nanoparticle (SNP) assay were all used. BChE anticholinesterase activity was determined using 5,50-dithiobis(2-nitrobenzoic) acid (DTNB). The  $\alpha$ -amylase inhibition activity of the iodine/potassium iodide (IKI) technique was used to modify the inhibitory activity of  $\alpha$ -amylase.

Antioxidant activity tests reveal that all extracts of the studied plants have antioxidant properties at varying levels. The IC<sub>50</sub> values include the aqueous fraction and the methanolic fraction of the ABTS radical with values IC<sub>50</sub> = (81.990.82; 95,233,54 mg/ml) respectively, cupric reducing the antioxidant capacity of the AG with an IC<sub>50</sub> value of 72.0815.21, and HL (257.3810.11). While that of phenanthrolines obtained with AG extract (33.91±12.60), HL extract showed a higher A<sub>0.5</sub> value (92.32±49.07).

A non aparent BChE inhibitory activity was demonstrated by both AJ and HL extracts (IC<sub>50</sub>>200). The investigation of the antidiabetic effect shows that HL extracts have amylase inhibitory activity (percent of inhibition=52.34).

Finally, aqueous extracts of *A. Judaica* and *H. lingulata* have significant antioxidant activity and anti-diabetic properties and can be exploited in food and pharmaceutical industries.

Key word : *A. judaica* ,*H. lingulata*, enzymatic inhibitory, Antioxidant activity

## Résumé

Les plantes médicinales sont considérées comme des sources naturelles d'antioxydants et jouent donc un rôle important dans la résolution des problèmes de santé et le traitement des maladies.

*Hyacinthoides lingulata* et *Artemisia Judaica* sont deux plantes médicinales qui ont longtemps été connues pour leurs propriétés médicinales. Nous avons testé l'activité antioxydante, l'activité inhibitrice de la cholinestérase et l'activité antidiabétique à l'aide d'extraits dans cette étude.

La macération est utilisée pour obtenir les extraits. Six méthodes ont été utilisées pour tester l'activité antioxydante des extraits : le DPPH free radical scavenging assay, le Reducing Power Assay, le ABTS Radical Cation Scavenging assay, phenanthroline assay, Cupric Reducing Antioxidant assay, et le silver nanoparticle (SNP) assay ont tous été utilisés.

L'activité anticholinestérasique du BChE a été déterminée à l'aide de l'acide 5,50-dithiobis(2-nitrobenzoïque) (DTNB).

L'inhibition  $\alpha$ -amylase de la technique iode/iodure de potassium (IKI) a été utilisée pour modifier l'activité inhibitrice de l' $\alpha$ -amylase.

Les tests d'activité antioxydante révèlent que tous les extraits des plantes étudiées ont des propriétés antioxydantes à différents niveaux. Les valeurs IC<sub>50</sub> incluent la fraction aqueuse et la fraction méthanolique du radical ABTS avec des valeurs IC<sub>50</sub> = (81.990.82; 95,233,54 mg/ml) respectivement, cuprique réduisant la capacité antioxydante de l'AG avec une valeur IC<sub>50</sub> de 72.0815.21, et HL (257.3810.11). Alors que celle des phénanthrolines obtenues avec l'extrait AG (33.91 12.60), l'extrait HL a montré une valeur A0.5 plus élevée (92.32 49.07). Une activité inhibitrice non aparente de la BChE a été démontrée par les extraits AJ et HL (IC<sub>50</sub>>200). L'étude de l'effet antidiabétique montre que les extraits de HL ont une activité inhibitrice de l'amylase (pourcentage d'inhibition = 52,34).

Enfin, les extraits aqueux d'*A. Judaica* et de *H. linguelata* ont une activité antioxydante et des propriétés antidiabétiques importantes et peuvent être exploités dans les industries alimentaires et pharmaceutiques.

Mot-clé : *A. judaica*, *H. lingulata*, enzymatique inhibitif, activité antioxydante



## ملخص

تُفهم الأعشاب الطبية على أنها مصادر طبيعية مضادة للأكسدة، وبالتالي تلعب دورًا مهمًا في حل المشكلات الصحية وعلاج الأمراض.

، هما نبتتان طبيبتان معروفتان منذ فترة طويلة بخصائصهما *Hyacinthoides lingulata* و *Artemisia Judaica* الطبية. اختبرنا النشاط المضاد للأكسدة، والنشاط المثبط للكولينستيراز، والنشاط المضاد للسكري باستخدام المستخلصات في هذه الدراسة.

يُستخدم التكاثر للحصول على المستخلصات. تم استخدام ست طرق لاختبار النشاط المضاد للأكسدة للمستخلصات: اختبار ، ومقاييس ABTS Radical Cation Scavenging ، ومقاييس الطاقة المنخفضة، ومقاييس DPPH زبال الجذور الحرة تم تحديد نشاط مضادات (SNP P) الفينانثرولين، ومقاييس مضادات الأكسدة المنخفضة للكوبريك، والجسيمة النانوية الفضية تم استخدام نشاط تثبيط (DTNB) (2-nitrobenzoic) dithiobis باستخدام حمض 5، BChE الكولين استراز. أميلاز- $\alpha$  لتعديل النشاط المثبط لـ (IKI) لتقنية اليود/يوديد البوتاسيوم  $\alpha$  الأميلاز.

تكشف اختبارات النشاط المضاد للأكسدة أن جميع المستخلصات النباتية المدروسة لها خصائص مضادة للأكسدة بمستويات  $IC_{50} = 95 \pm 81.990.82$  الجذري مع القيم ABTS الكسر المائي وجزء الميثانوليك من  $IC_{50}$  متفاوتة. تشمل قيم HL من 72,0815,21، و  $IC_{50}$  بقيمة AG 54 233 (ملغم/مل)، على التوالي، تقل من قدرة مضاد الأكسدة في AG ( $33.91 \pm$ ) في حين أن مستخلص الفينانثرولين الذي تم الحصول عليه باستخدام مستخلص (257,3810,11) ( $92.32 \pm 49.07$ ) أعلى A0.5 قيمة HL ، أظهر مستخلص ( $12.60$ ).

يظهر التحقيق في (HL  $IC_{50} > 200$ ) و AJ بواسطة كل من مستخلصات BChE تم إثبات نشاط مثبط غير باري لها نشاط مثبط للأميلاز (نسبة مئوية من التثبيط = 52.34) HL التأثير المضاد للسكري أن مستخلصات

بنشاط كبير مضاد للأكسدة وخصائص مضادة *H. lingulata* و *A. Judaica* أخيرًا، تتمتع المستخلصات المائية من لمرض السكري ويمكن استغلالها في الصناعات الغذائية والصيدلانية.

**الكلمات المفتاحية** *H. lingulata*, *A. judaica*, BChE ، المضاد للسكري ، مضادات الأكسدة ، المستخلصات النباتية

**Année universitaire : 2021-2022**

**Présenté par :**

Aissaoui Roumayssa, Messili Dounia,  
Taleb Oussama Rami

**Biological activities of *Artemisia Judaica* and *Hyacinthoides lingulata***

Medicinal plants are understood as natural antioxidant sources, and thus play an important role in resolving health issues and treating diseases. *Hyacinthoides lingulata* and *Artemisia Judaica*, are two medicinal plants that have long been known for their medicinal properties. We tested antioxidant activity, cholinesterase inhibitory activity, and anti-diabetic activity using extracts in this study. Maceration is used to obtain the extracts. Six methods were used to test the antioxidant activity of the extracts : DPPH free radical scavenging assay, the Reducing Power Assay, ABTS Radical Cation Scavenging assay, phenanthroline assay, Cupric Reducing Antioxidant assay, and the silver nanoparticle (SNP) assay were all used. BChE anticholinesterase activity was determined using 5,50-dithiobis(2-nitrobenzoic) acid (DTNB). The  $\alpha$ -amylase inhibition activity of the iodine/potassium iodide (IKI) technique was used to modify the inhibitory activity of  $\alpha$ -amylase. Antioxidant activity tests reveal that all extracts of the studied plants have antioxidant properties at varying levels. The IC<sub>50</sub> values include the aqueous fraction and the methanolic fraction of the ABTS radical with values IC<sub>50</sub> = (81.990.82; 95,233,54 mg/ml) respectively, cupric reducing the antioxidant capacity of the AG with an IC<sub>50</sub> value of 72.0815.21, and HL (257.3810.11). While that of phenanthrolines obtained with AG extract (33.91±12.60), HL extract showed a higher A0.5 value (92.32±49.07). A non aparent BChE inhibitory activity was demonstrated by both AJ and HL extracts (IC<sub>50</sub>>200). The investigation of the antidiabetic effect shows that HL extracts have amylase inhibitory activity (percent of inhibition=52.34). Finally, aqueous extracts of *A. Judaica* and *H. lingulata* have significant antioxidant activity and anti-diabetic properties and can be exploited in food and pharmaceutical industries.

**Mots-clefs :** *A. judaica*, *H. lingulata*, enzymatic inhibitory, antioxidant activity

**Laboratoires de recherche :**

Université Frères Mentouri, Constantine 1.

**Encadreur :** RAMLI Iman (MAA- Université Frères Mentouri, Constantine 1).

**Examineur 1 :** HADDAD Souad MAA - Université Frères Mentouri, Constantine 1).

**Examineur 2 :** BENLATRECHE Moufida (MAA - Université Frères Mentouri, Constantine 1).